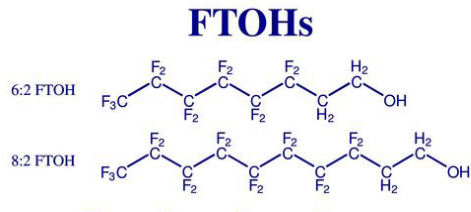


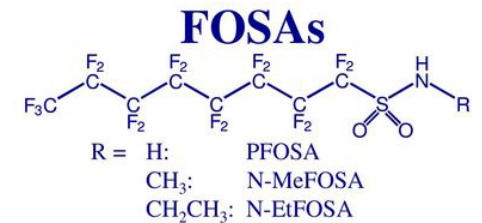
# Global presence and fate of PFAS precursors



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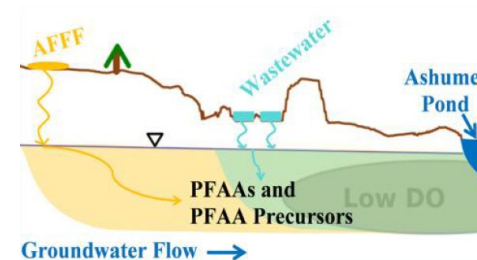
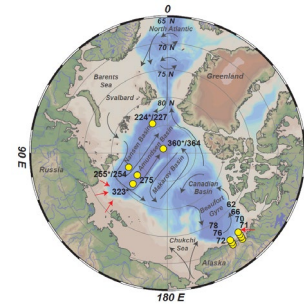
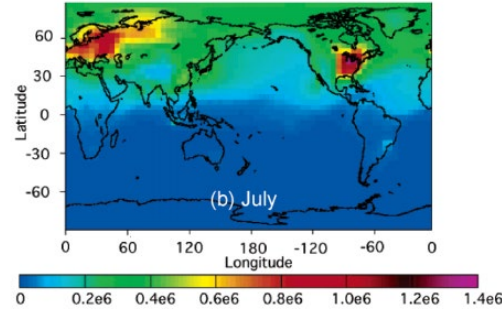
 **NIH**  
NIEHS

STEEP is funded by the Superfund Research Program, National Institute of Environmental Health Sciences under award number P42ES027726.  
More information about STEEP is available at: <http://web.uri.edu/steep/> and [https://tools.niehs.nih.gov/erp/programs/Program\\_detail.cfm?Project\\_ID=P42ES027726](https://tools.niehs.nih.gov/erp/programs/Program_detail.cfm?Project_ID=P42ES027726)

 **STEEP**  
Sources, Transport, Exposure & Effects of PFASs  
UNIVERSITY OF RHODE ISLAND SUPERFUND RESEARCH PROGRAM

# Presence of atmospheric FTOHs

- GLOBAL presence of FTOHs.
- Precursors affecting the Arctic and foodwebs
- Precursors in plants
- Microbial conversions of precursors

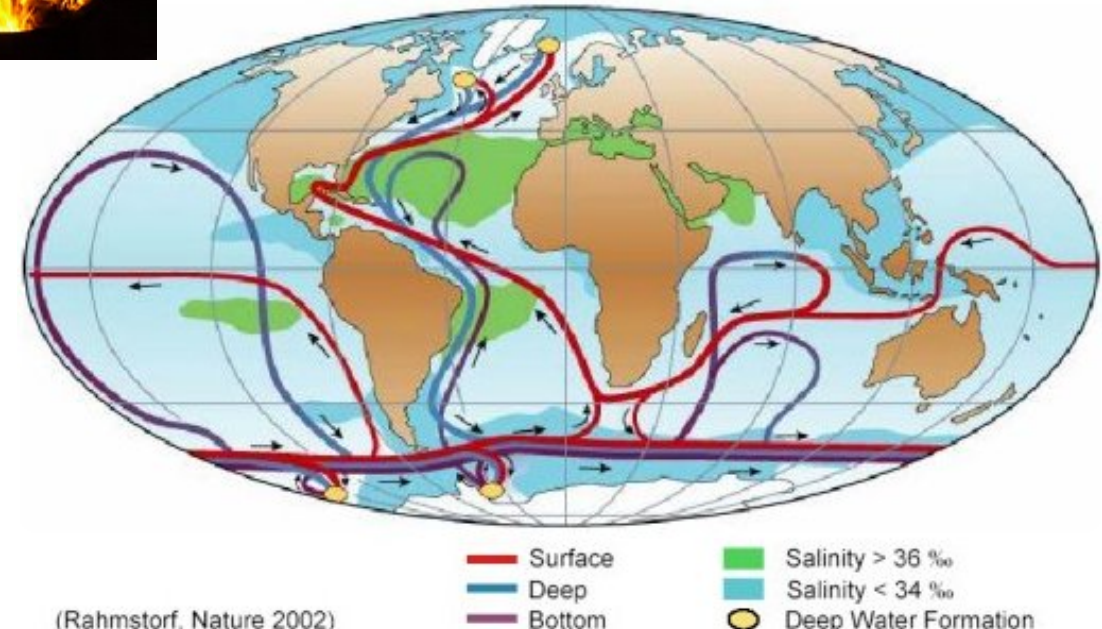
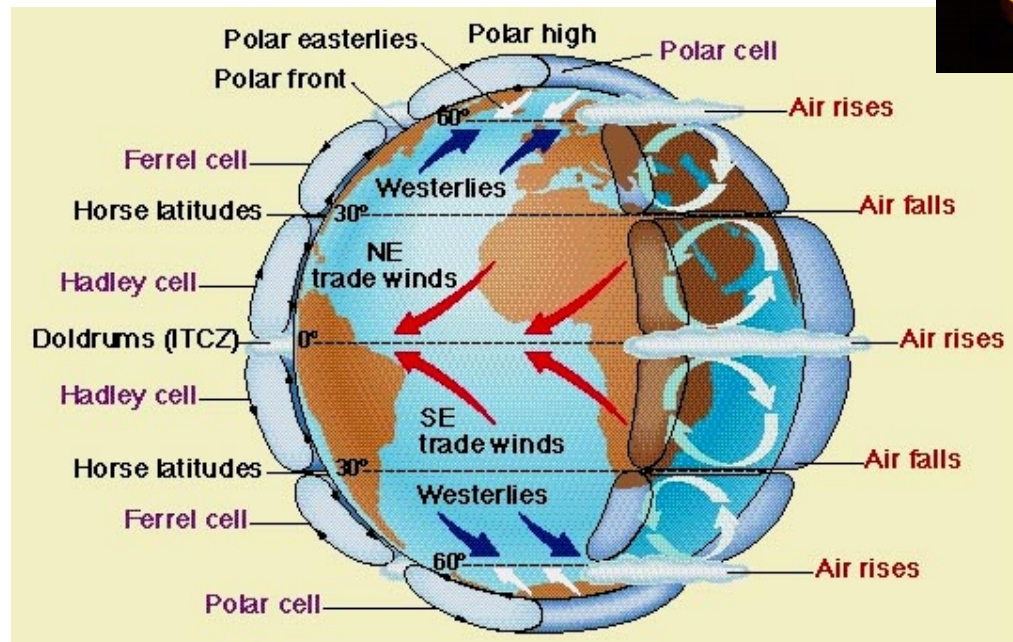


# PFAS battle of the elements:

## Winds

## vs

## Currents



<http://www.oceansonline.com/winds.htm>



# Volatile PFAS in the atmosphere: FTOHs

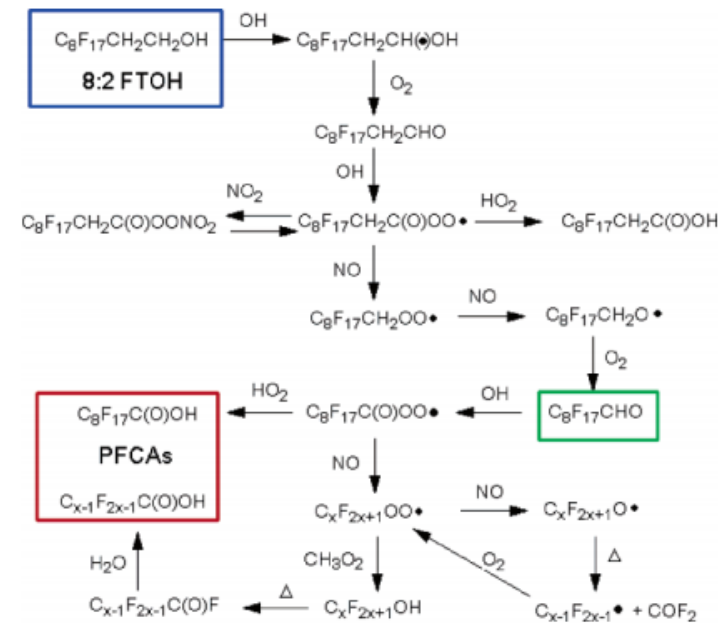
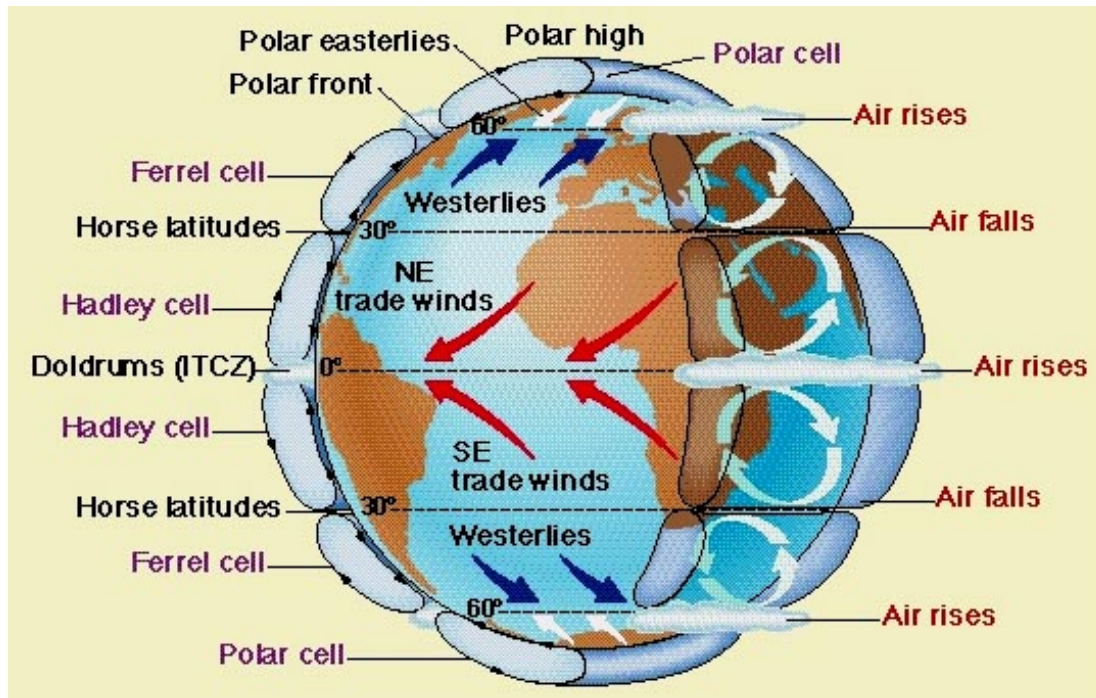
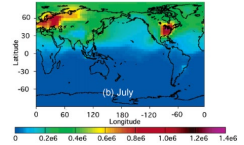


FIGURE 1. Simplified mechanism for the atmospheric degradation of 8:2 FTOH (blue box) illustrating its conversion into  $\text{C}_8\text{F}_{17}\text{CHO}$  (green box) and the competition between NO and either  $\text{HO}_2$  or  $\text{CH}_3\text{O}_2$  radicals that limits the formation of perfluorocarboxylic acids (red box).

(Wallington et al., 2006)

# Time for radical ideas

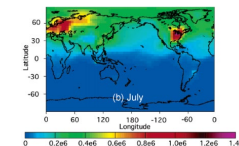


TABLE 1. Atmospheric Oxidation Mechanism of 8:2 FTOH

reaction	rate
<b>Atmospheric Chemistry of C<sub>8</sub>F<sub>17</sub>CH<sub>2</sub>CH<sub>2</sub>OH</b>	
C <sub>8</sub> F <sub>17</sub> CH <sub>2</sub> CH <sub>2</sub> OH + OH → C <sub>8</sub> F <sub>17</sub> CH <sub>2</sub> CHO	3.2 × 10 <sup>-11</sup> exp(-1000/T) <sup>a</sup>
<b>Atmospheric Chemistry of C<sub>8</sub>F<sub>17</sub>CH<sub>2</sub>CHO</b>	
C <sub>8</sub> F <sub>17</sub> CH <sub>2</sub> CHO + OH → C <sub>8</sub> F <sub>17</sub> CH <sub>2</sub> C(O)OO	1.0 × 10 <sup>-10</sup> exp(-1000/T) <sup>a</sup>
C <sub>8</sub> F <sub>17</sub> CH <sub>2</sub> C(O)OO + NO → C <sub>8</sub> F <sub>17</sub> CHO	8.1 × 10 <sup>-12</sup> exp(270/T) <sup>a</sup>
C <sub>8</sub> F <sub>17</sub> CH <sub>2</sub> C(O)OO + NO <sub>2</sub> → C <sub>8</sub> F <sub>17</sub> CH <sub>2</sub> C(O)OONO <sub>2</sub>	1.1 × 10 <sup>-11</sup> (T/298) <sup>-1.0 a</sup>
C <sub>8</sub> F <sub>17</sub> CH <sub>2</sub> C(O)OONO <sub>2</sub> → C <sub>8</sub> F <sub>17</sub> CH <sub>2</sub> C(O)OO + NO <sub>2</sub>	2.8 × 10 <sup>16</sup> exp(-13580/T) <sup>b</sup>
C <sub>8</sub> F <sub>17</sub> CH <sub>2</sub> C(O)OO + HO <sub>2</sub> → products	0.6 × 4.3 × 10 <sup>-13</sup> exp(1040/T) <sup>a</sup>
C <sub>8</sub> F <sub>17</sub> CH <sub>2</sub> C(O)OO + HO <sub>2</sub> → C <sub>8</sub> F <sub>17</sub> CHO + CO <sub>2</sub>	0.4 × 4.3 × 10 <sup>-13</sup> exp(1040/T) <sup>a</sup>
<b>Atmospheric Chemistry of C<sub>8</sub>F<sub>17</sub>CHO</b>	
C <sub>8</sub> F <sub>17</sub> CHO + OH → C <sub>8</sub> F <sub>17</sub> C(O)OO	1.7 × 10 <sup>-11</sup> exp(-1000/T) <sup>a</sup>
C <sub>8</sub> F <sub>17</sub> CHO + hν → C <sub>8</sub> F <sub>17</sub> O <sub>2</sub>	φ = 0.02
C <sub>8</sub> F <sub>17</sub> C(O)OO + NO → C <sub>8</sub> F <sub>17</sub> O <sub>2</sub>	8.1 × 10 <sup>-12</sup> exp(270/T) <sup>a</sup>
C <sub>8</sub> F <sub>17</sub> C(O)OO + NO <sub>2</sub> → C <sub>8</sub> F <sub>17</sub> C(O)OONO <sub>2</sub>	1.1 × 10 <sup>-11</sup> (T/298) <sup>-1.0 a</sup>
C <sub>8</sub> F <sub>17</sub> C(O)OONO <sub>2</sub> → C <sub>8</sub> F <sub>17</sub> C(O)OO + NO <sub>2</sub>	2.8 × 10 <sup>16</sup> exp(-13580/T) <sup>b</sup>
C <sub>8</sub> F <sub>17</sub> C(O)OO + HO <sub>2</sub> → C <sub>8</sub> F <sub>17</sub> COOH(PFNA <sup>c</sup> ) + O <sub>3</sub>	0.10 × 4.3 × 10 <sup>-13</sup> exp(1040/T) <sup>a</sup>
C <sub>8</sub> F <sub>17</sub> C(O)OO + HO <sub>2</sub> → C <sub>8</sub> F <sub>17</sub> O <sub>2</sub>	0.90 × 4.3 × 10 <sup>-13</sup> exp(1040/T) <sup>a</sup>
<b>Atmospheric Chemistry of C<sub>8</sub>F<sub>17</sub>O<sub>2</sub></b>	
C <sub>8</sub> F <sub>17</sub> O <sub>2</sub> + NO → C <sub>8</sub> F <sub>17</sub> O + NO <sub>2</sub>	2.8 × 10 <sup>-12</sup> exp(300/T) <sup>a</sup>
C <sub>8</sub> F <sub>17</sub> O <sub>2</sub> + HO <sub>2</sub> → C <sub>8</sub> F <sub>17</sub> O + OH + O <sub>2</sub>	4.1 × 10 <sup>-13</sup> exp(750/T) <sup>a</sup>
C <sub>8</sub> F <sub>17</sub> O <sub>2</sub> + CH <sub>3</sub> O <sub>2</sub> → C <sub>8</sub> F <sub>17</sub> O + CH <sub>3</sub> O	2.7 × 10 <sup>-12</sup> exp(-470/T) <sup>a</sup>
C <sub>8</sub> F <sub>17</sub> O <sub>2</sub> + CH <sub>3</sub> O <sub>2</sub> → C <sub>8</sub> F <sub>17</sub> OH + HCHO	1.0 × 10 <sup>-13</sup> exp(660/T) <sup>a</sup>
C <sub>8</sub> F <sub>17</sub> OH → C <sub>7</sub> F <sub>15</sub> COOH(PFOA <sup>d</sup> )	2.3 × 10 <sup>-6 b</sup>

<sup>a</sup> Units of cm<sup>3</sup> molecule<sup>-1</sup> s<sup>-1</sup>. <sup>b</sup> Units of s<sup>-1</sup>. <sup>c</sup> PFNA = perfluorononanoic acid = C<sub>8</sub>F<sub>17</sub>COOH. <sup>d</sup> PFOA = perfluorooctanoic acid = C<sub>7</sub>F<sub>15</sub>COOH.

(Wallington et al., 2006)

# The spread of FTOHs across the atmosphere

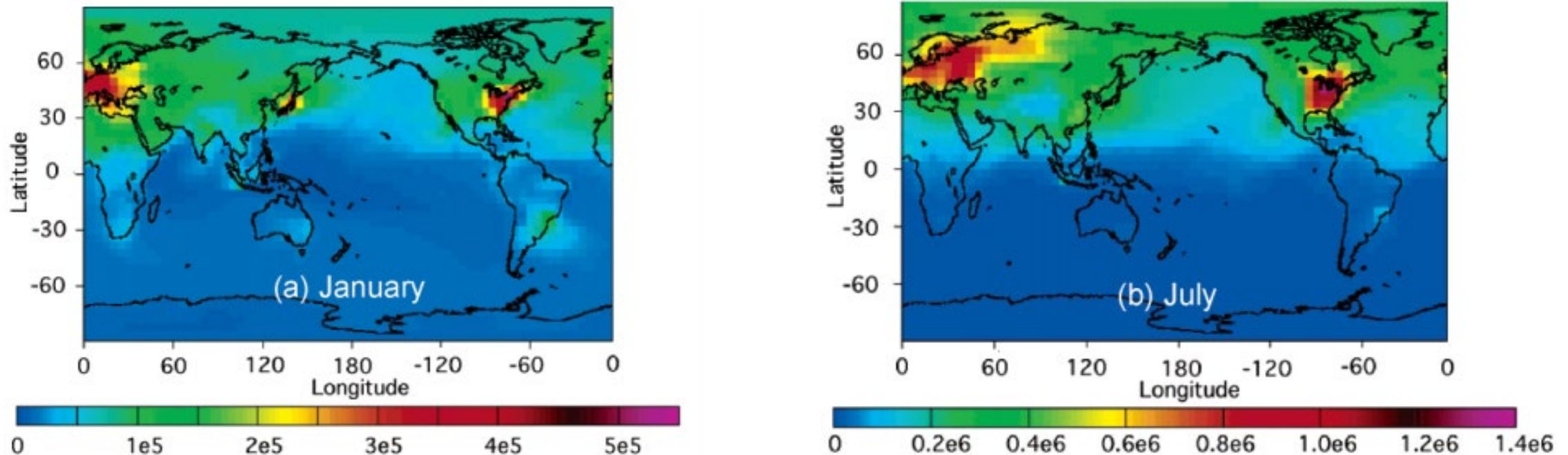
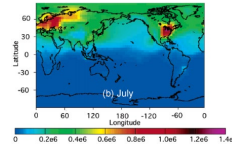


FIGURE 2. Summed concentration of 8:2 FTOH and all of its degradation products at 50 m in altitude for (a) January and (b) July. The color scale extends from (a) 0 to  $5.5 \times 10^5$  molecule  $\text{cm}^{-3}$  and (b) 0 to  $1.4 \times 10^6$  molecule  $\text{cm}^{-3}$ .

- Yield of 3-6% PFOA from 8:2 FT-OH; also smaller PFCAs
- Source: Arctic 5:1
- $t_{\text{life}} \sim 20\text{-}40$  days (Wallington et al., 2006)

# Toronto: Arctic 5:1, first set

Toronto: Costa Rica: Botswana 10:5:1

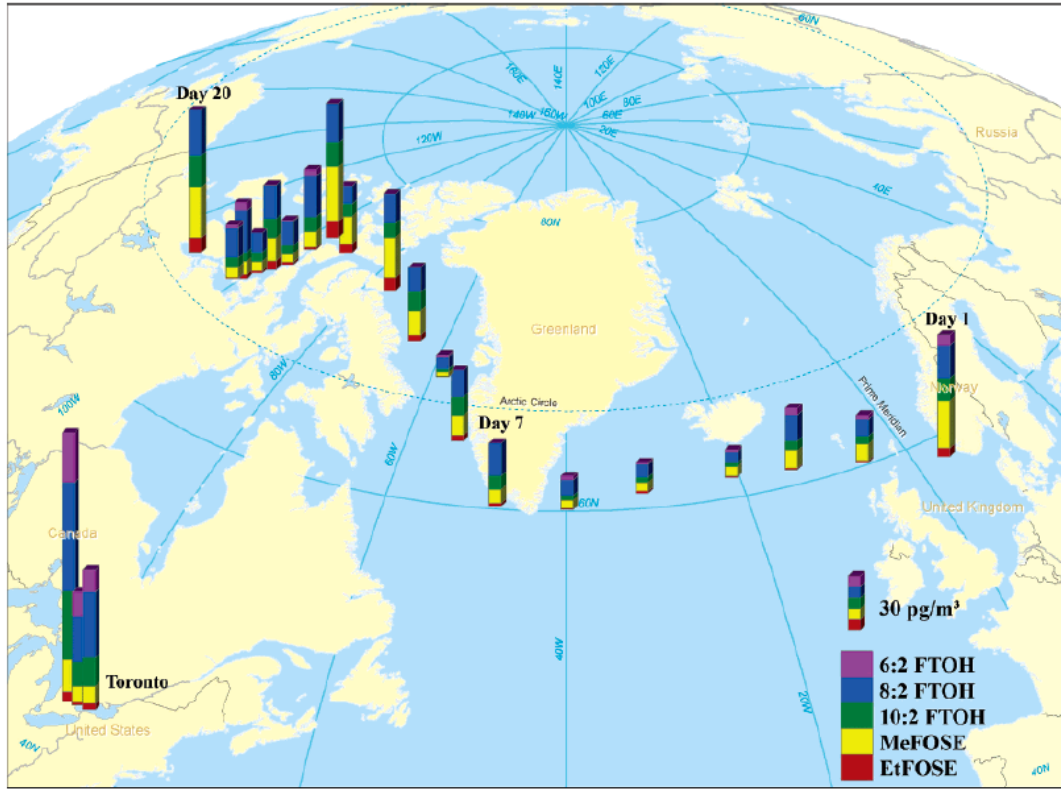
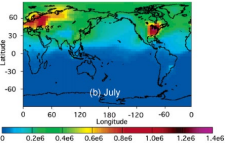
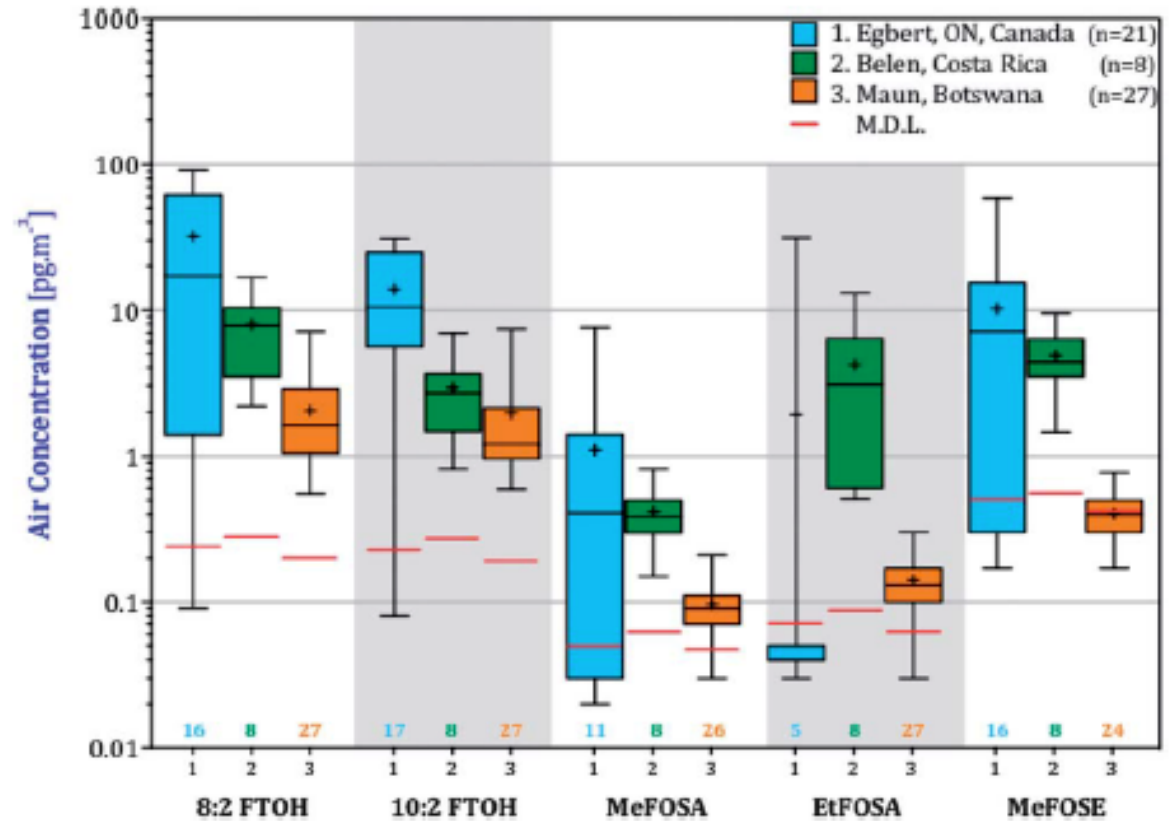


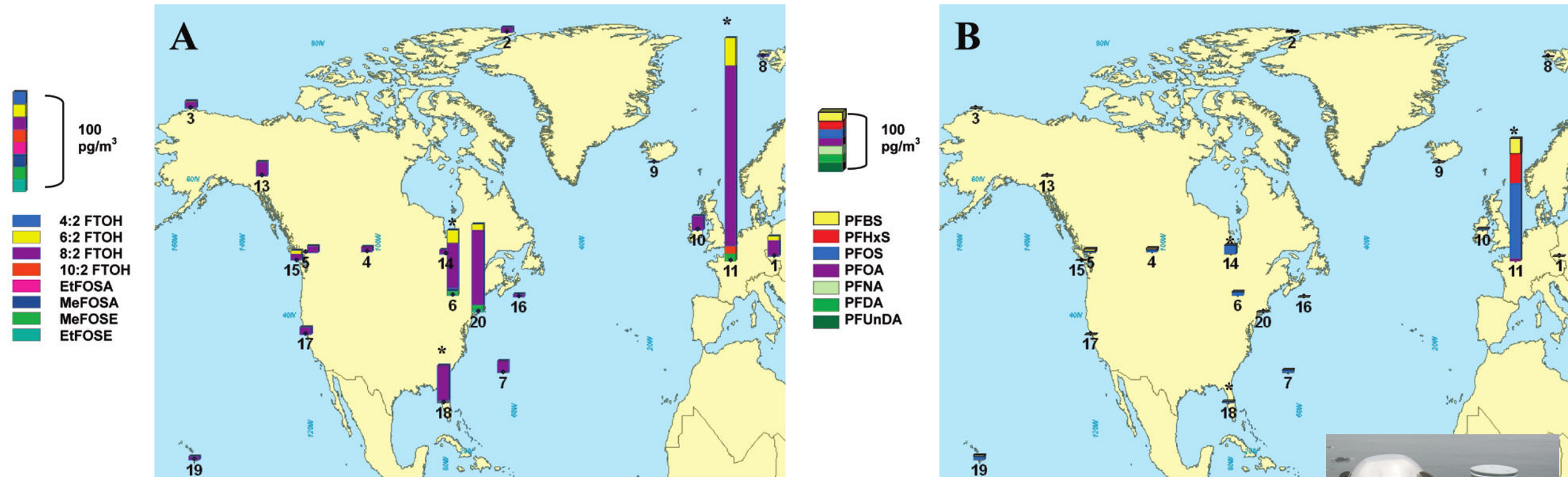
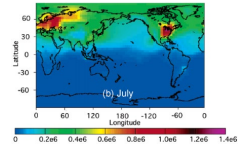
FIGURE 2. Total air concentrations (sum of gas phase and particle phase) for FTOHs and PFASs across the North Atlantic Ocean at Canadian Archipelago (see Figure 1 cruise track) and in Toronto, Canada.

(Shoeib et al., 2006)



(Gawor et al., 2014)

# Some sites are cleaner



(Genualdi et al., 2010)





# All clean down south?

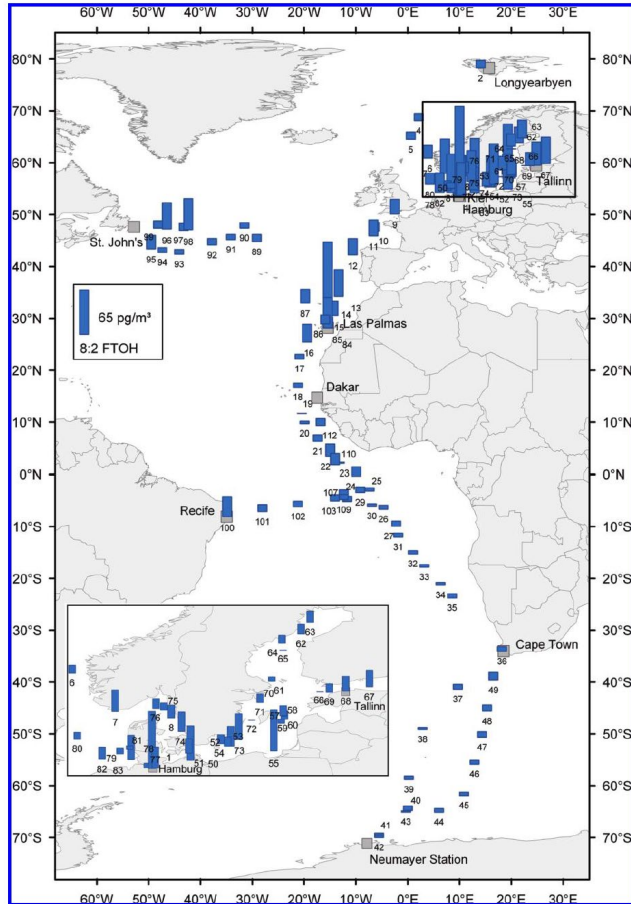
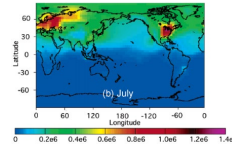
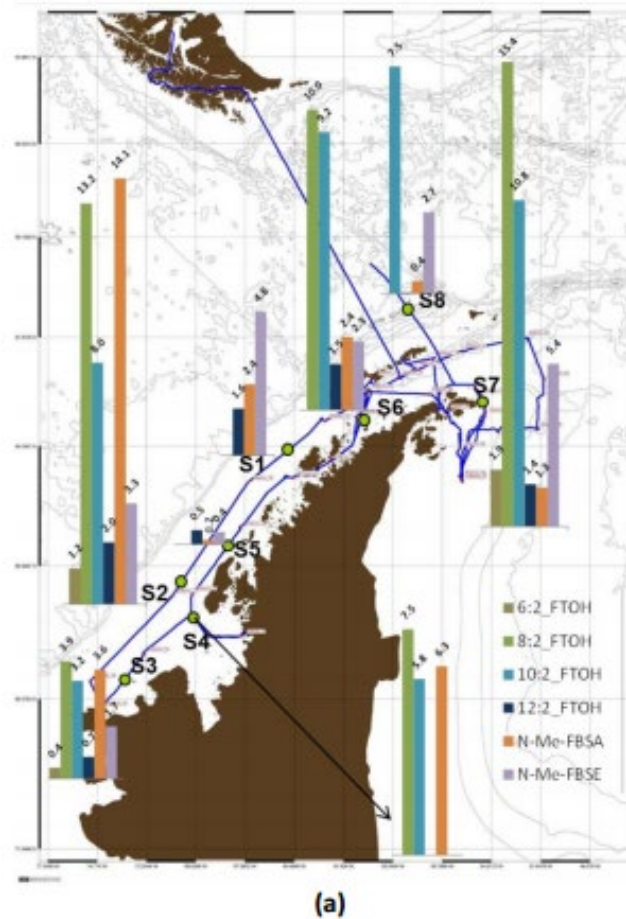


FIGURE 1. Spatial distribution of 8:2 FTOH gas-phase concentrations determined during several cruises in the Baltic Sea and the Atlantic and Southern Oceans. Note that the close-up of the Baltic Sea region is not to scale.

(Dreyer et al., 2009)



(Del Vento et al., 2012)

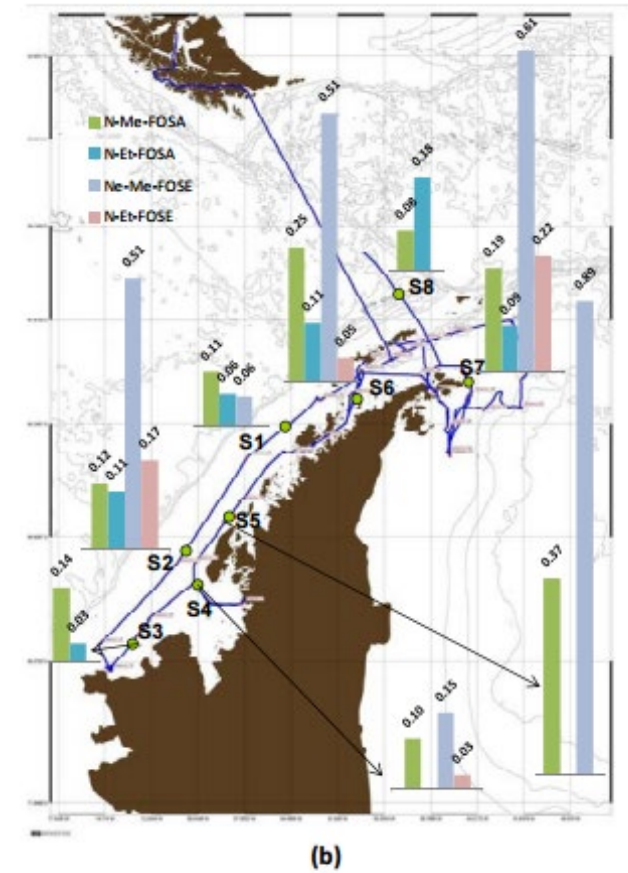


Figure 1. Gaseous concentrations ( $\text{pg m}^{-3}$ ) of FTOHs, MeFBSA and MeFBSE (a); FOSAs and FOSEs (b), measured during February 2009.

# Cold condensation or fata morgana?

- Resurgence of FTOHs in Antarctic?
- Trends across Atlantic Ocean as expected
- Revolatilization of FTOHs from snow?
- Yet
  - NH 4x more land
  - NH 9 x more people

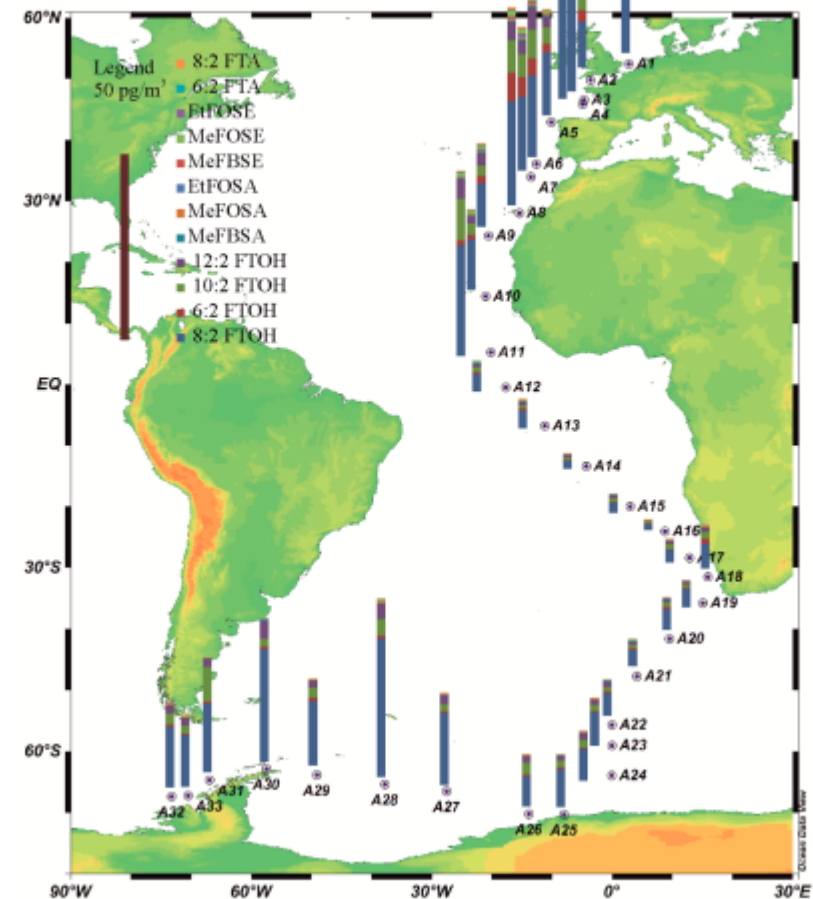
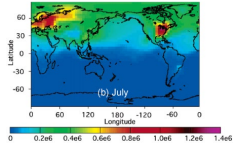
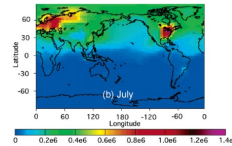


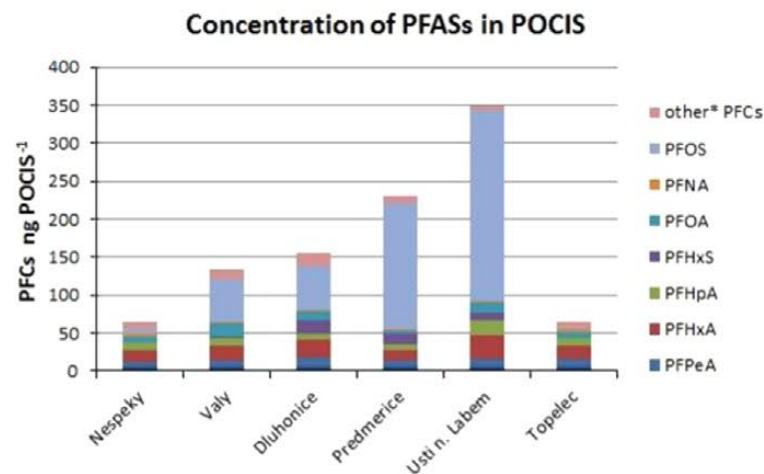
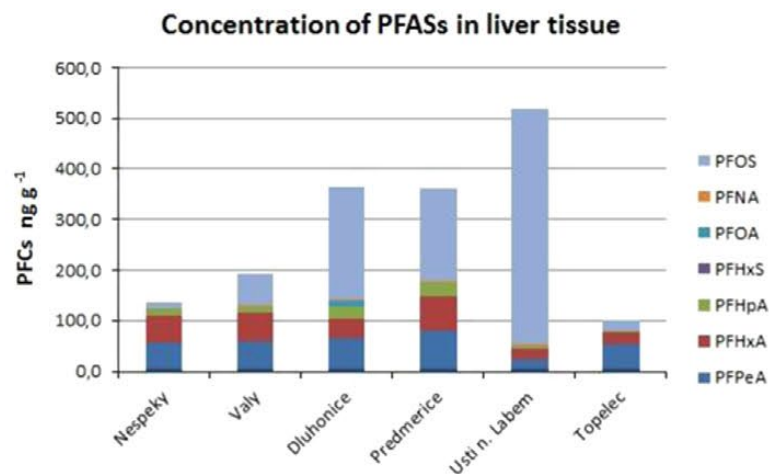
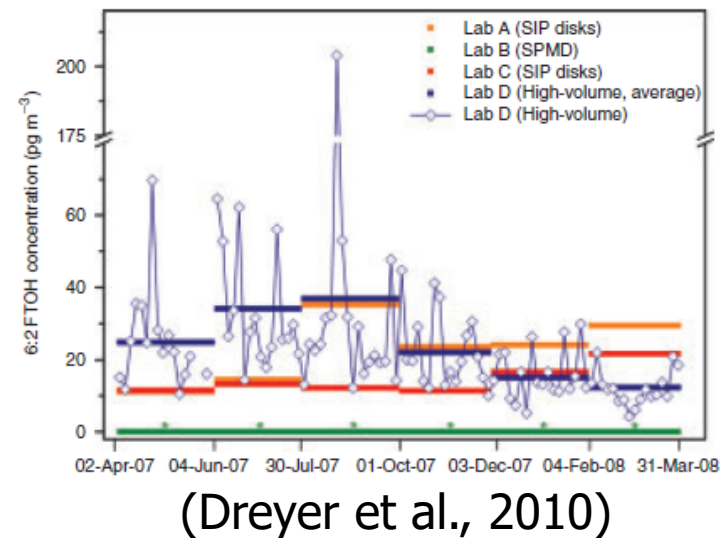
Figure 1. High volume air sampling sites during the cruise of the R/V *Polarstern* in ANT-XXII from October 2010 to January 2011 from Germany to the Antarctic, and concentrations ( $\text{pg}/\text{m}^3$ ), composition of individual PFASs in the marine atmosphere.

(Wang et al., 2015)



# Why passives might be useful

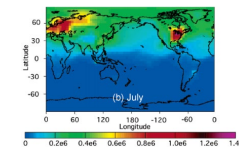
- Time-integrated concentrations
- Pre-concentrations already in field
- Less handling in laboratory
- Proxy for bioaccumulation



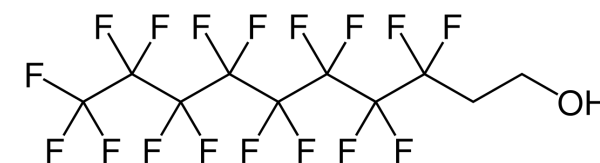
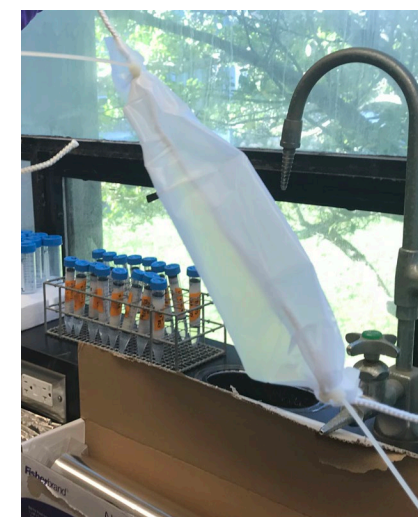
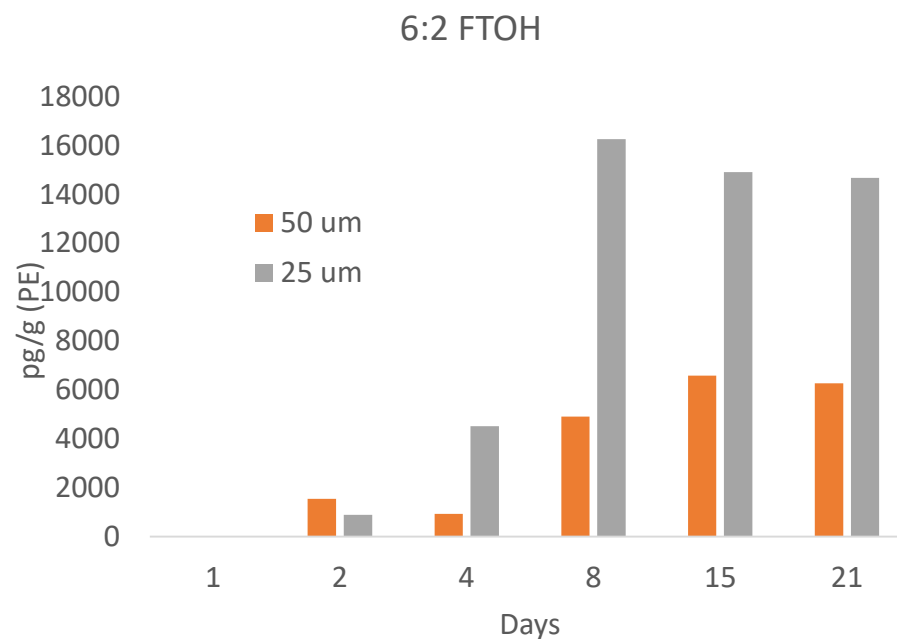
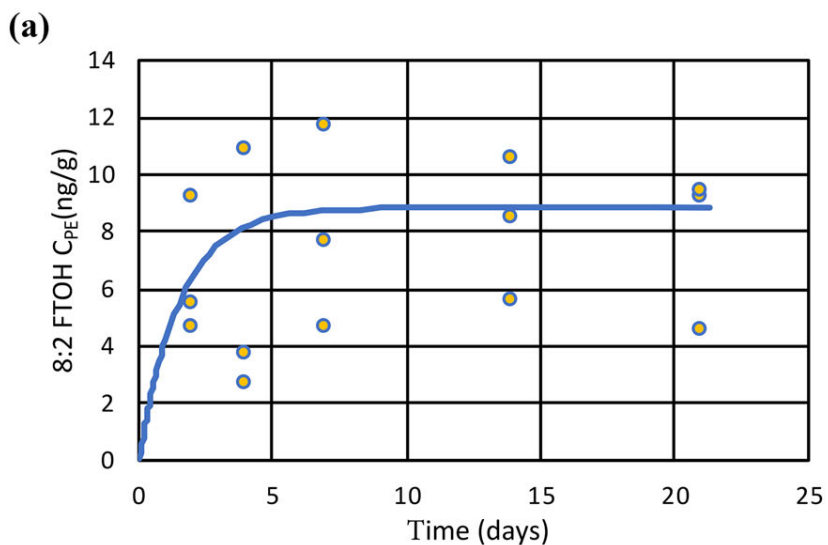
(Cervený et al., 2016)



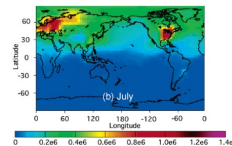
# Testing of Polyethylene Sheets as Passive Samplers for Volatile PFAS in Indoor Air



Maya Morales-McDevitt et al.



(Dixon-Anderson and Lohmann, 2018)



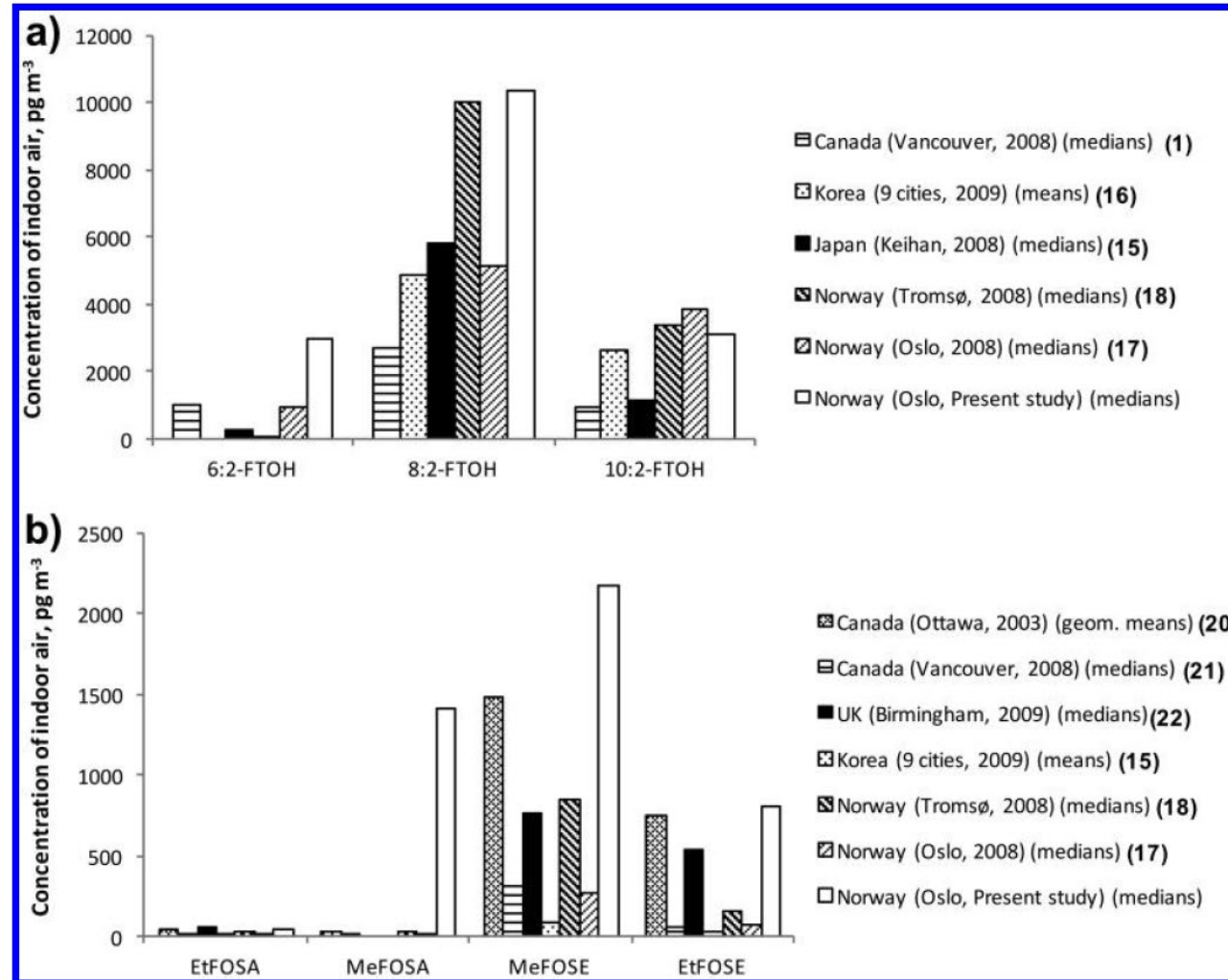
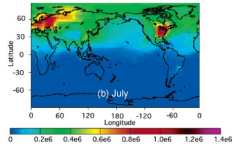
# Plenty of other precursors out there (AFFF)

Table 2. Product FTOH Concentrations and Nontargeted Chemical Compositions

product	targeted (ng $\mu\text{L}^{-1}$ )				nontargeted		
	4:2 FTOH	6:2 FTOH	8:2 FTOH	10:2 FTOH	composition	peak center $m/Q$	mass error (ppm)
Masurf FS-115	26.1	578.5	106.8	43.0	$\text{C}_6\text{F}_{10}\text{H}_2\text{O}$	279.996	5.0
					$\text{C}_7\text{F}_{13}\text{H}_3\text{O}$	349.994	10.4
					$\text{C}_{16}\text{F}_{26}\text{H}_8\text{O}$	710.009	9.9
					$\text{C}_{18}\text{F}_{30}\text{H}_8\text{O}$	810.002	9.4
					$\text{C}_{20}\text{F}_{34}\text{H}_8\text{O}$	909.988	16.7
Zonyl FSA	9.4	948.1	130.7	17.9	$\text{C}_{14}\text{F}_{23}\text{H}_5\text{O}$	626.002	7.5
					$\text{C}_{16}\text{F}_{27}\text{H}_5\text{O}$	725.999	11.1
					$\text{C}_{18}\text{F}_{31}\text{H}_5\text{O}$	825.997	14.8
Capstone FS-35	6.7	644.6	–	–	$\text{C}_4\text{F}_5\text{H}_5\text{O}$	164.025	6.4
					$\text{C}_5\text{F}_7\text{H}_5\text{O}$	214.023	0.6
					$\text{C}_7\text{F}_{11}\text{H}_5\text{O}$	314.015	4.7
					$\text{C}_{14}\text{F}_{24}\text{H}_6\text{O}$	645.996	11.7
Arctic 3 AFFF	–	1.6	0.3	–	$\text{C}_7\text{F}_{11}\text{H}_5\text{O}$	314.018	4.9
					$\text{C}_{16}\text{F}_{27}\text{H}_5\text{O}$	726.002	15.3
					$\text{C}_{18}\text{F}_{31}\text{H}_5\text{O}$	825.986	1.8

(Riedel et al, 2019)

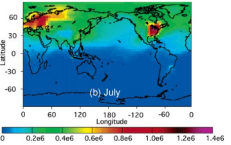
# Plenty of other precursors in there



(Padilla-Sanchez et al, 2017)

# Something in the air..

# ... and in your blood



Airborne Precursors Predict Maternal Serum Perfluoroalkyl Acid Concentrations

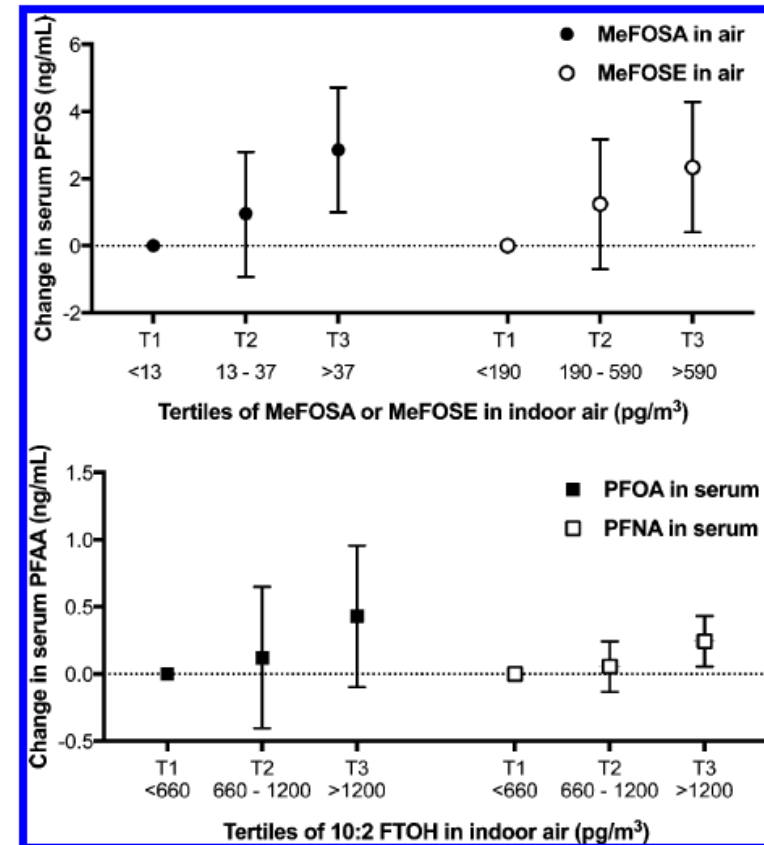
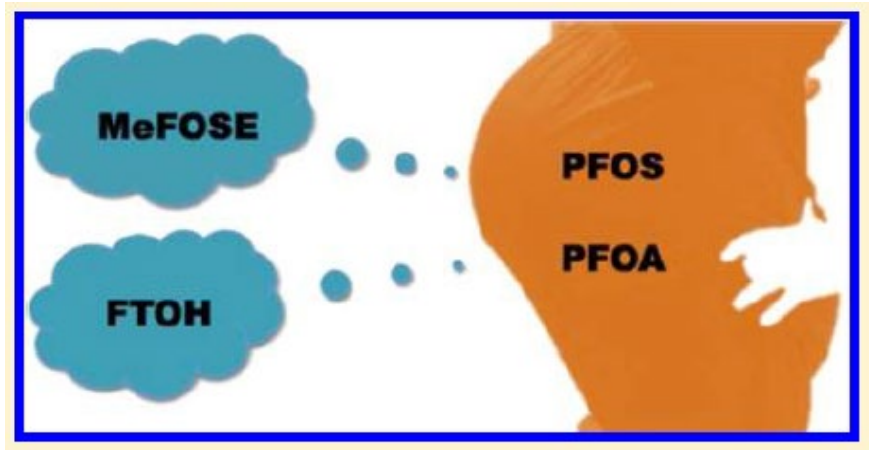
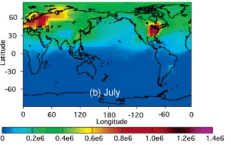


Figure 1. Exposure categories [low ( $n = 17$ ), medium ( $n = 16$ ), and high ( $n = 16$ )] of PreFAA in air predicting serum PFAA levels in 50 women.

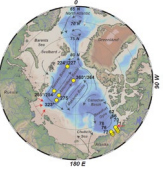
(Makey et al. 2017)

# Global presence –FTOHs

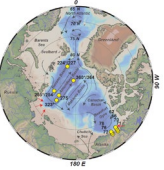


- Different campaigns – routine detection in NH;
- Gradients from source to sink regions
- Much stronger gradients indoor vs outdoor
- Passive samplers as 1 tool for routine detection
- Presence of long-chain acids in humans (pre-cursors)?

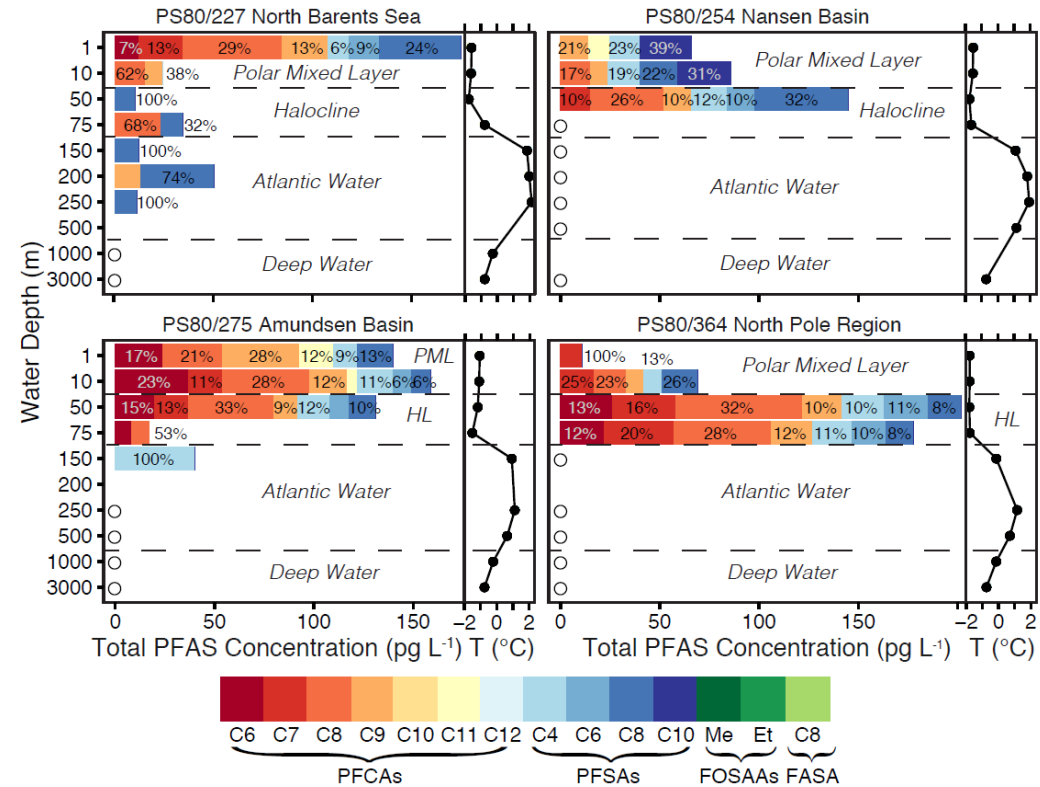
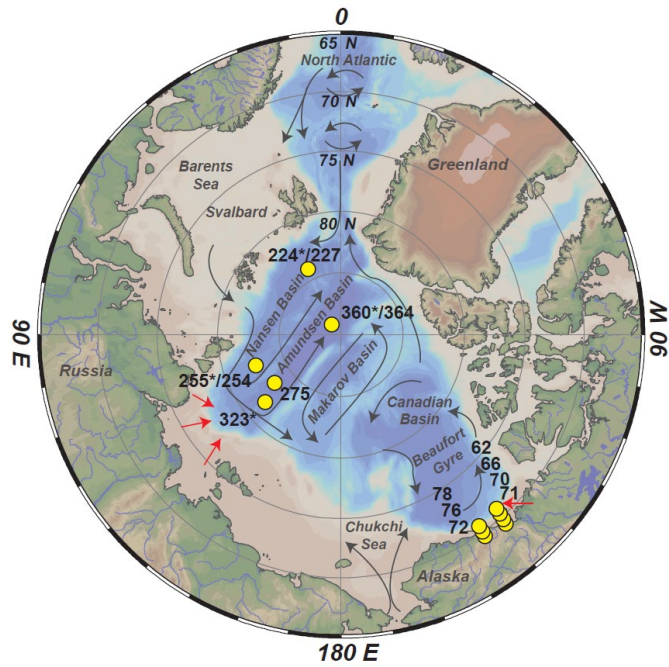




# Precursors into the Arctic



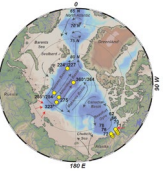
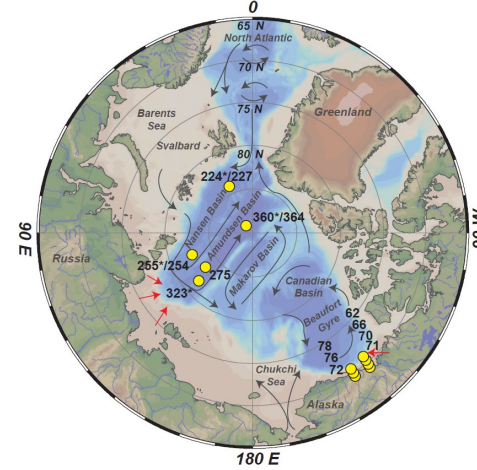
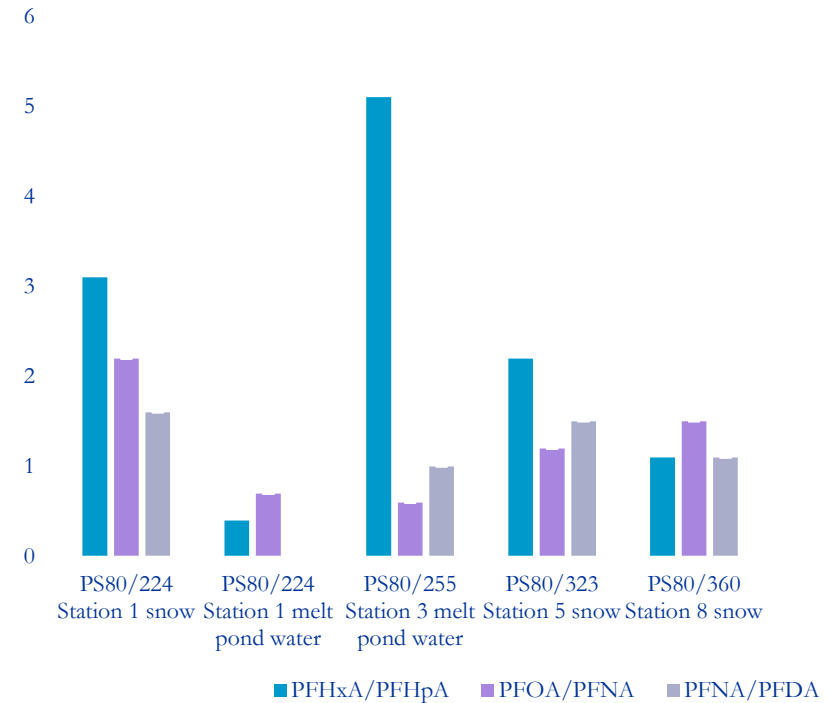
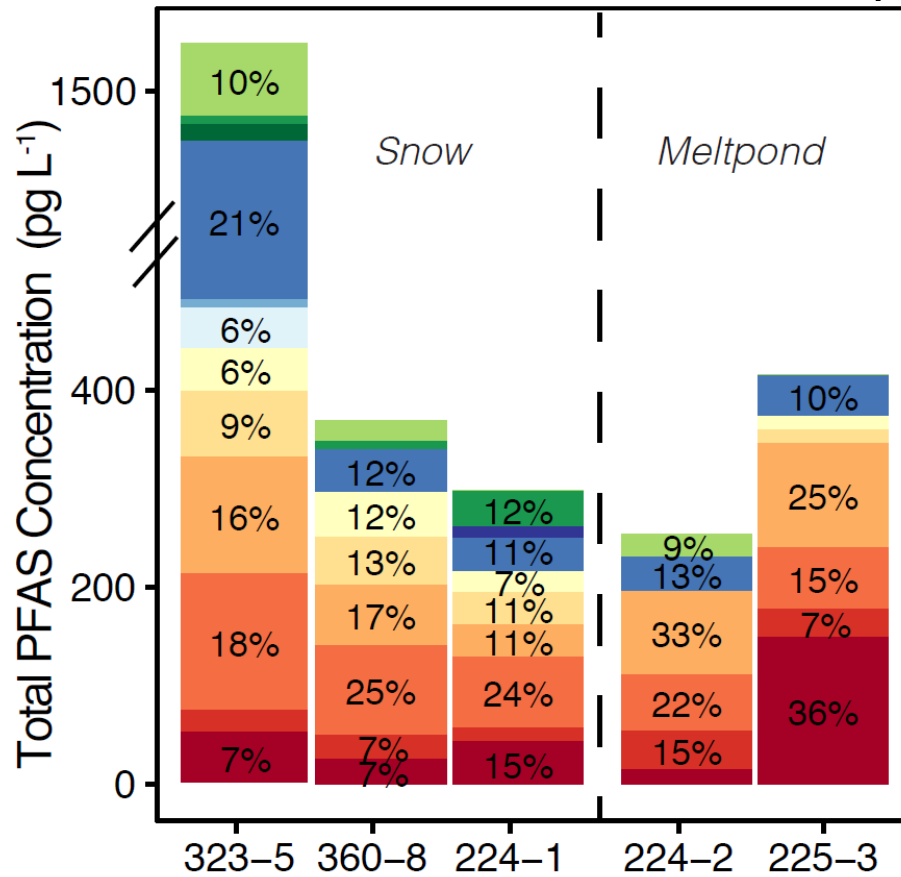
# First signs of Atlantic PFASs?



(Yeung et al., EST 2017)

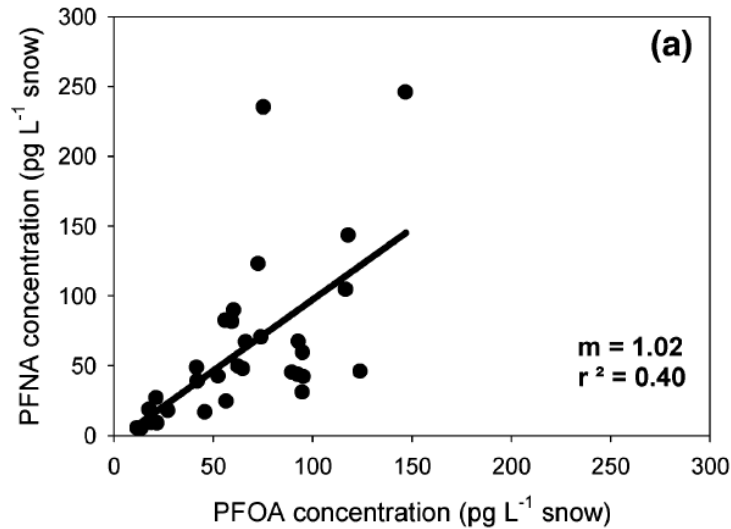
# Origin of PFASs on ice?

- Similar ratios on ice and in snow: deposition!

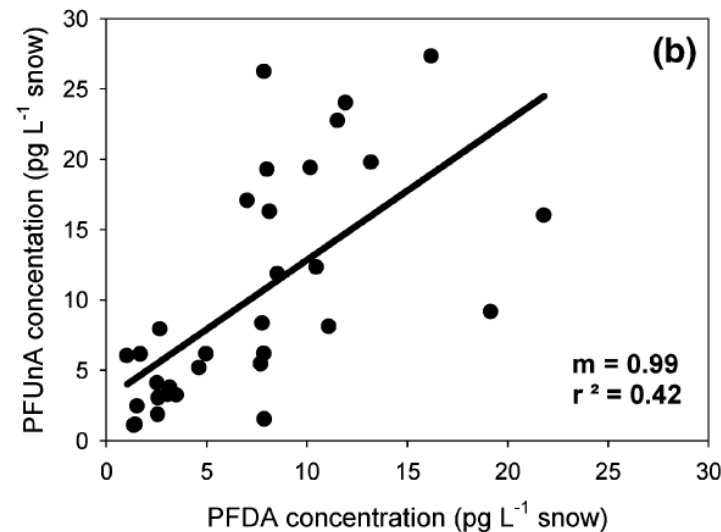
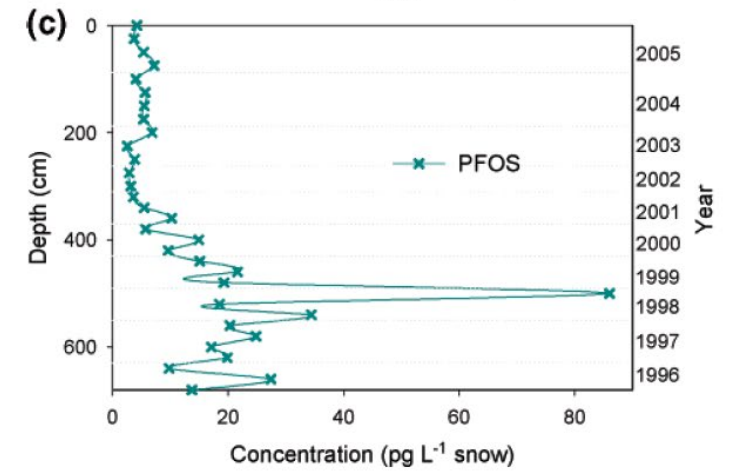
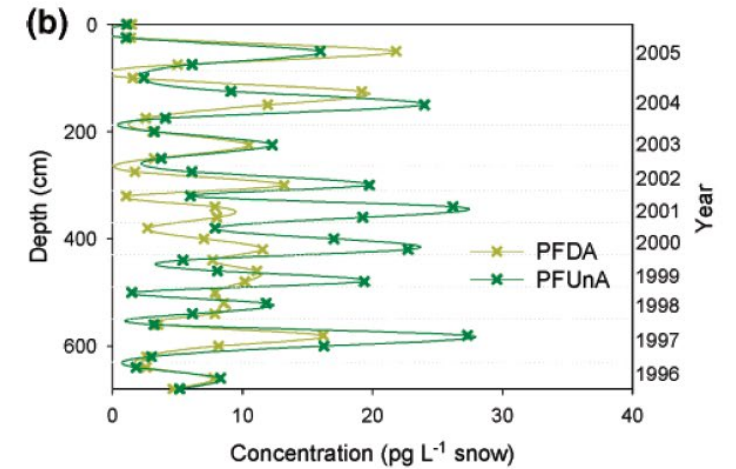
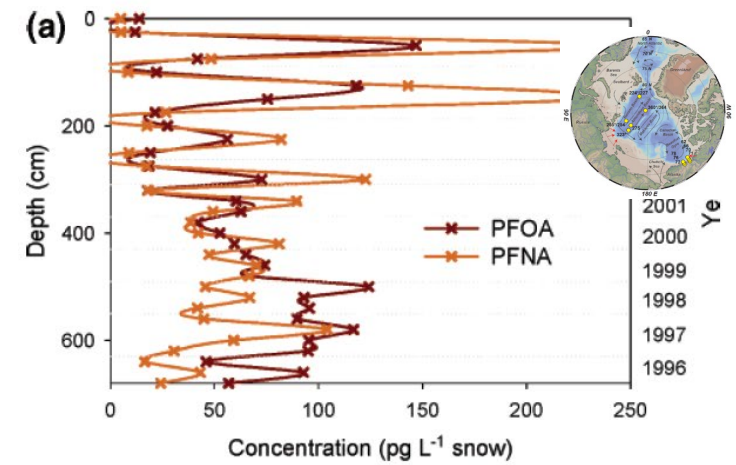


(Young et al., EST 2007)

# PFASs in Devon Ice Cap



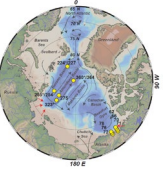
- Only atmospheric input
- Little from sea spray



- Oxidation of 8:2 FTOH  
→ PFOA, PFNA

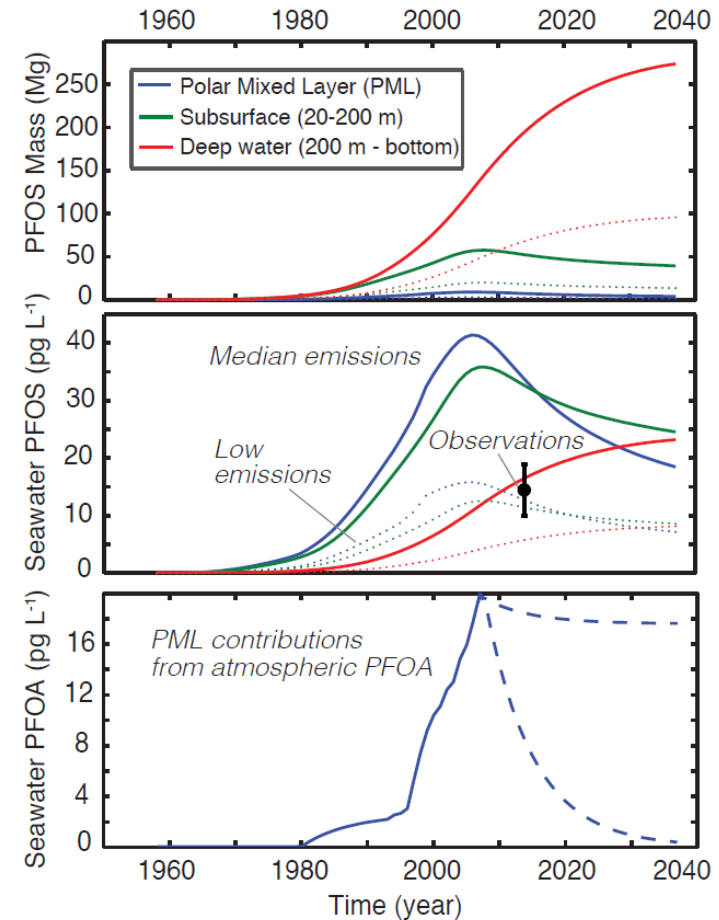
- Oxidation of 10:2 FTOH  
→ PFDA, PFUnDA  
(Little commercial production)

(Young et al., 2007)



# Back in the Arctic

- Only traces of PFASs in deeper Arctic water (Atlantic water mass)
- Little evidence for vertical transport
- Model suggest PFASs in deeper Arctic will increase
- Both atmosphere and ocean transport
  - For PFOS up to 30% from air
  - For PFOA 30-60% from air



(Zhang et al., GBC 2017)

# Bioaccumulation of PFASs in the Arctic

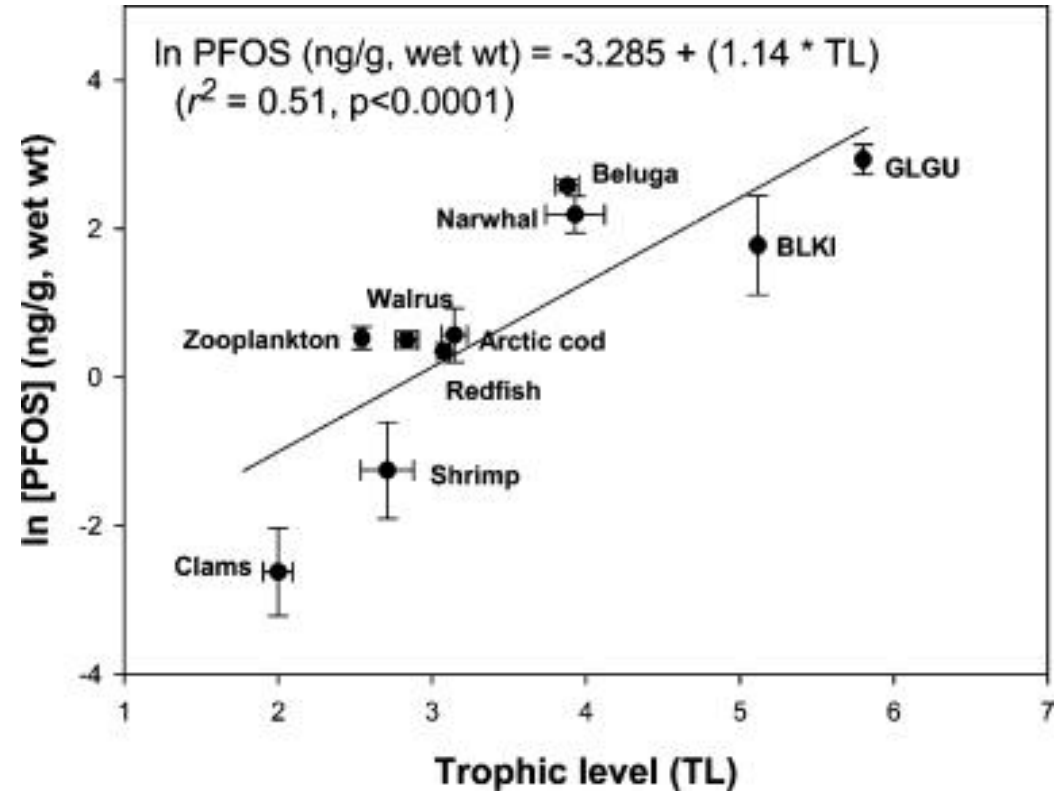
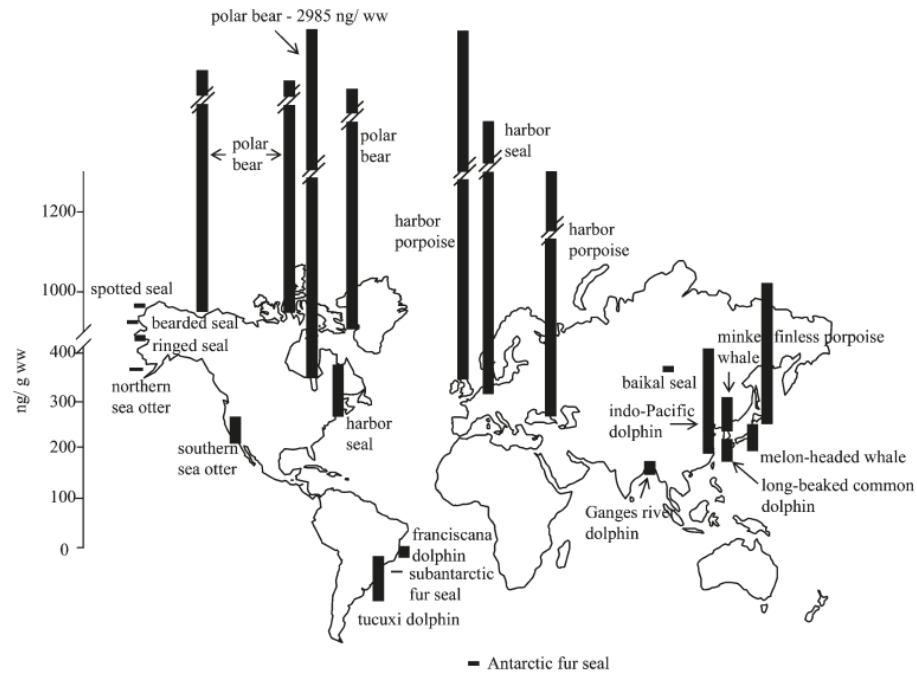
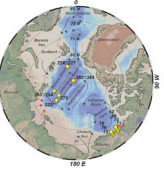


Figure 1. Recently reported (2006–2010) PFOS concentrations in liver of marine mammals worldwide.<sup>6–8,10,12,13,15,19,21,30,36,65–67,102–104</sup>

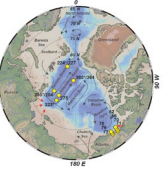
Monitoring of Perfluorinated Compounds in Aquatic Biota: An Updated Review

*PFCs in Aquatic Biota*

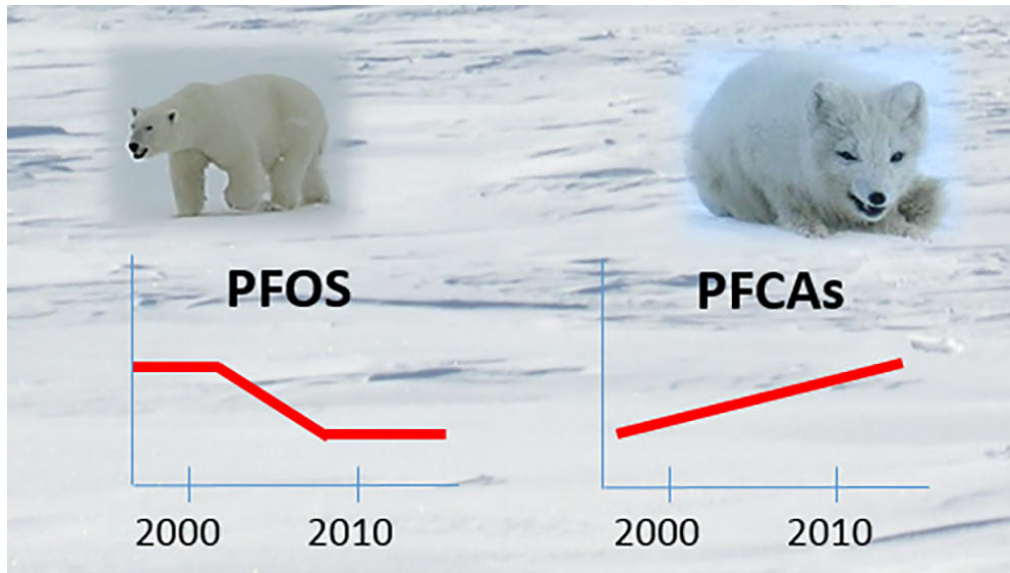
Magali Houde,<sup>\*,†</sup> Amila O. De Silva,<sup>‡</sup> Derek C. G. Muir,<sup>‡</sup> and Robert J. Letcher<sup>§</sup>

(Tomy et al., 2004)

# Foodweb effects in Arctic Ocean: PFOS



- Foodweb matters.



(Routti et al., 2017)

Rat~polar bear>seal>>>beluga

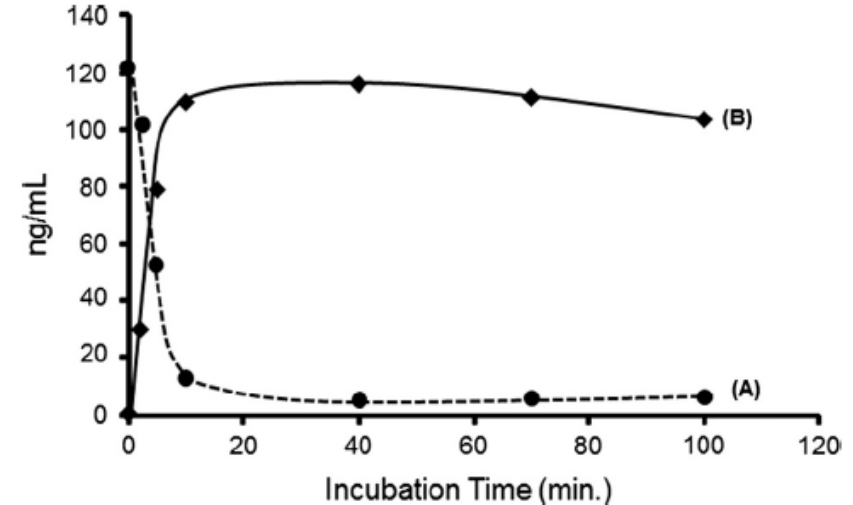
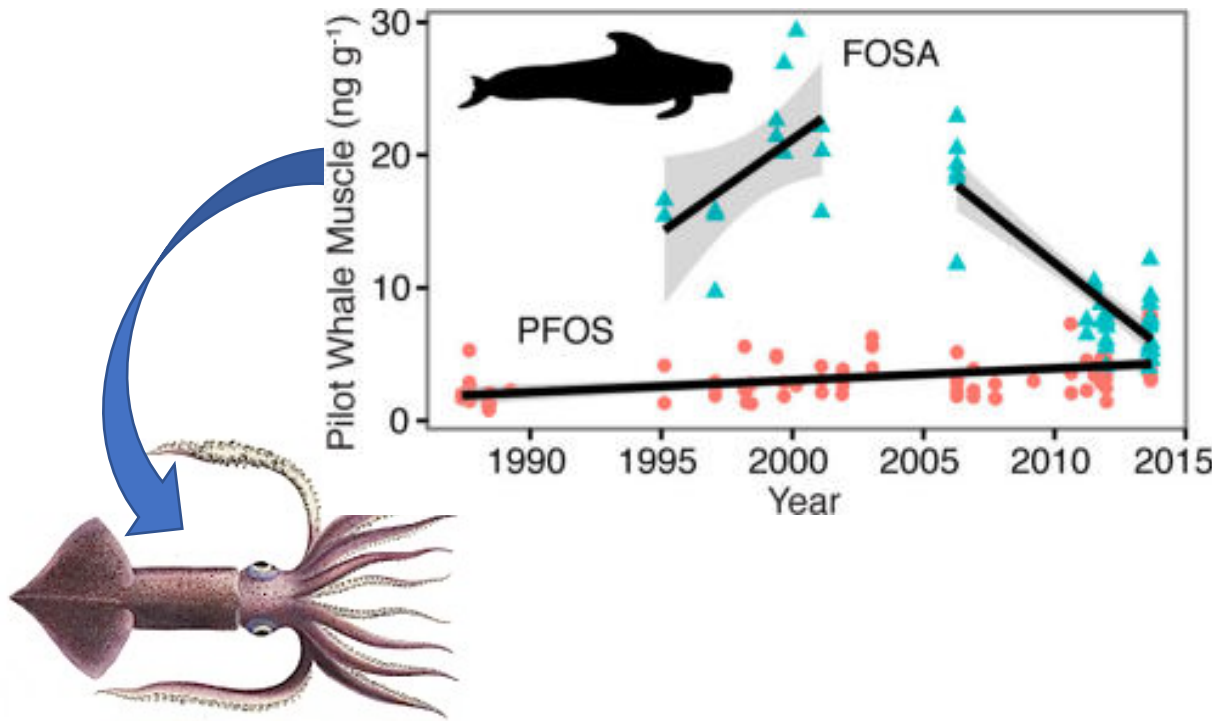
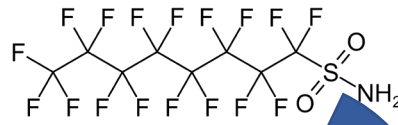


Fig. 1. Time course incubation of (A) *N*-EtFOSA depleted and (B) FOSA formed concentrations (based on  $n = 3$  replicate assays for each time point) in an *in vitro* liver assay using adult male Wistar-Han rat microsomes. The initial *N*-EtFOSA concentration was 300 nM.

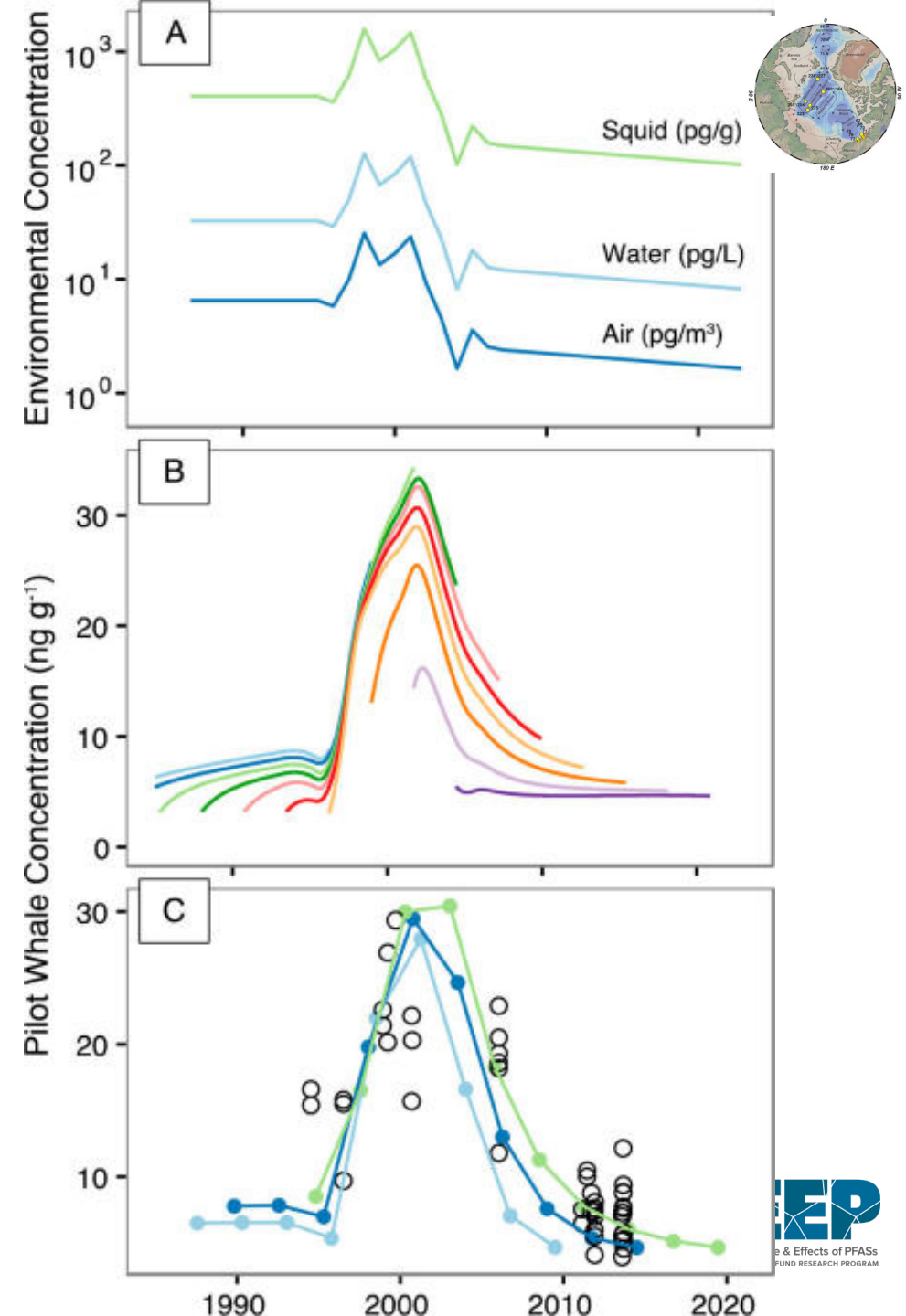
(Letcher et al., 2014)

# Pilot whales and FOSA

- Stay clear of metabolism



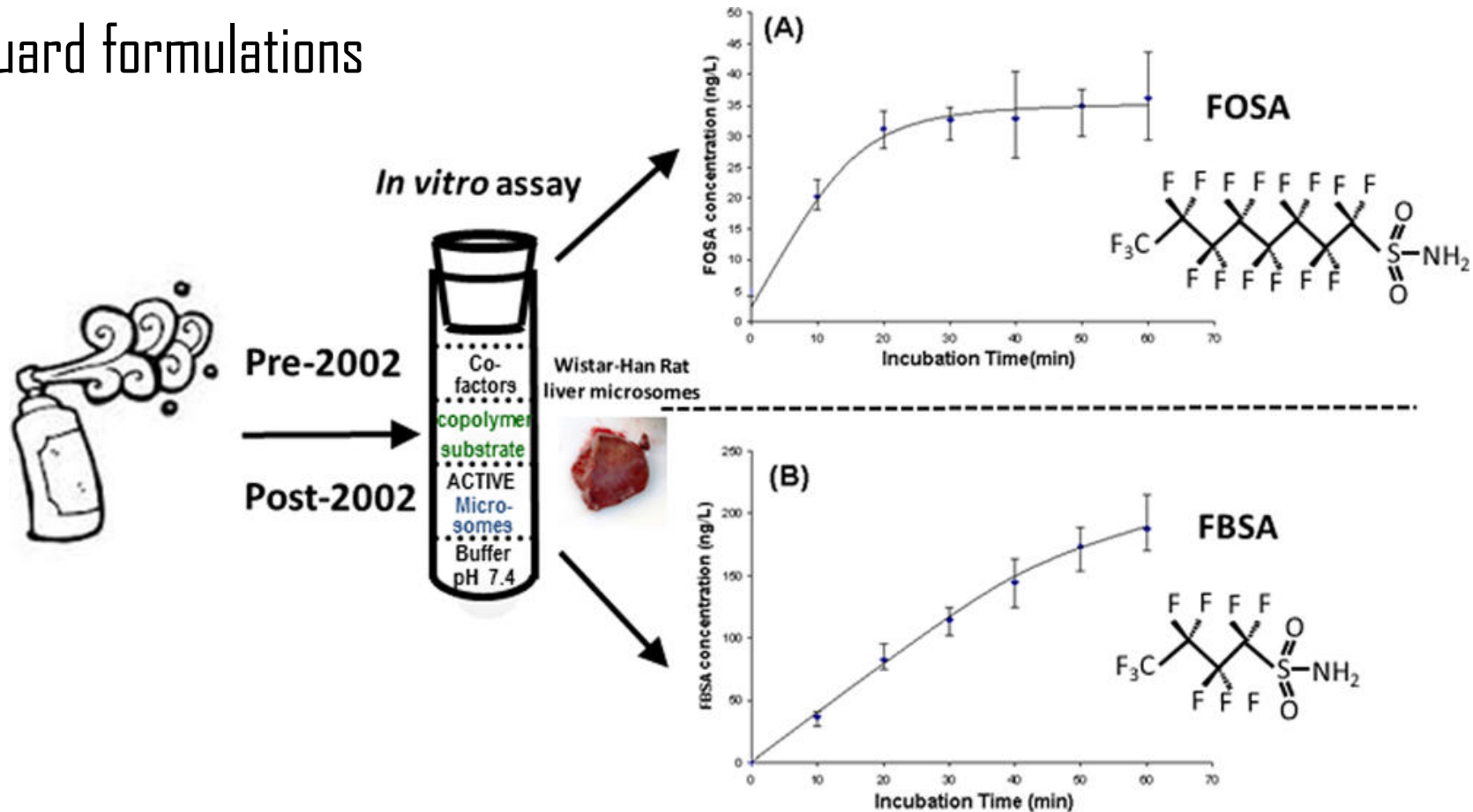
(Dassuncao et al., 2017)





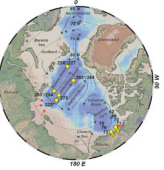
# FOSA is out.. FBSA is in..

- Scotchguard formulations



(Chu and Letcher, 2014)

# Fish bioaccumulation is a concern C6/ onwards



- PFAAs are well understood

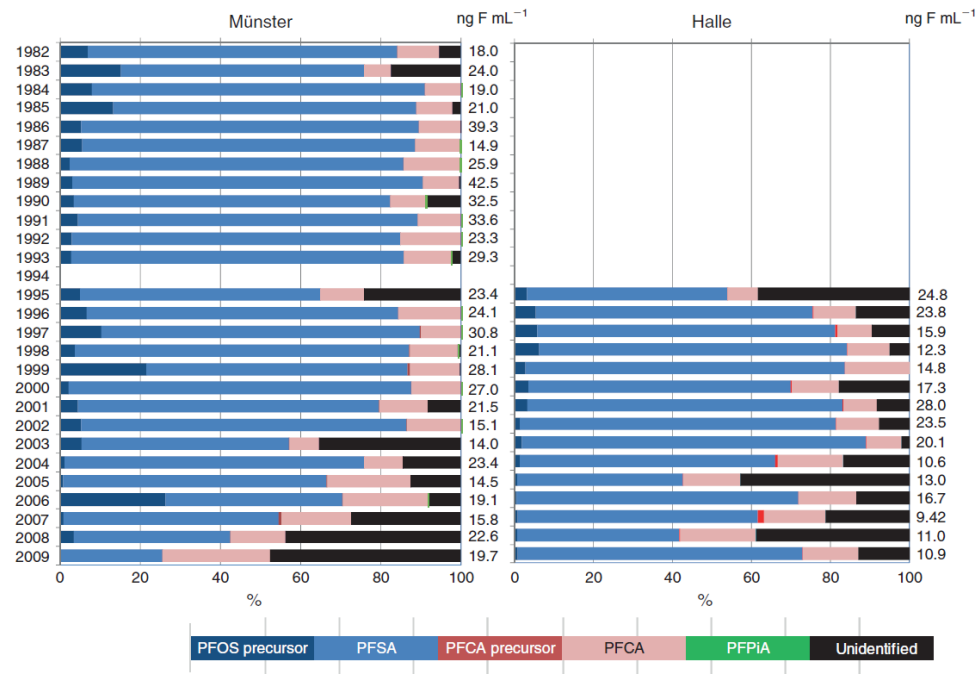
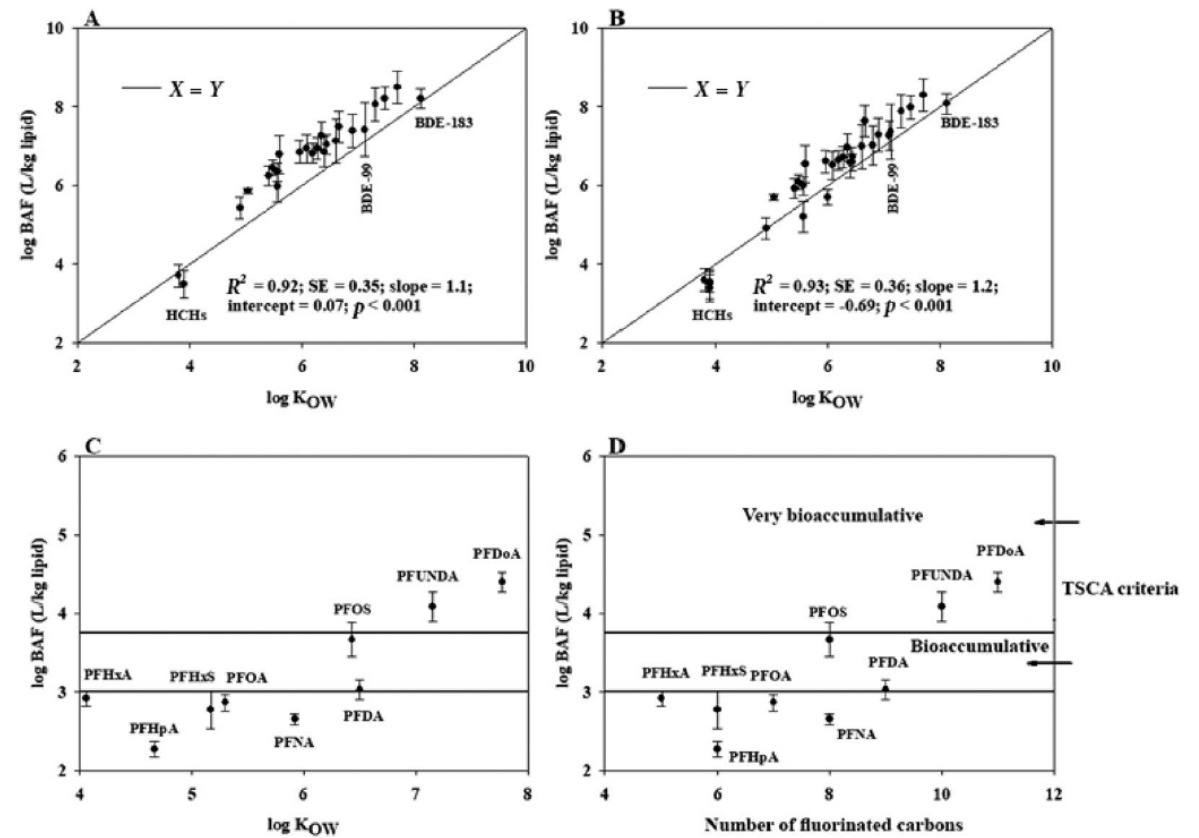


Fig. 4. Composition and concentrations (ng F mL<sup>-1</sup>) of extractable organofluorine (EOF) in German blood plasma samples (perfluorooctane sulfonate, PFOS; perfluoroalkyl sulfonate, PFSA; perfluorinated carboxylates, PFCAs; perfluorinated phosphinates, PFPIAs).

(Yeung and Mabury, 2016)



(Khairy et al., 2019)

# Precursors in the Arctic

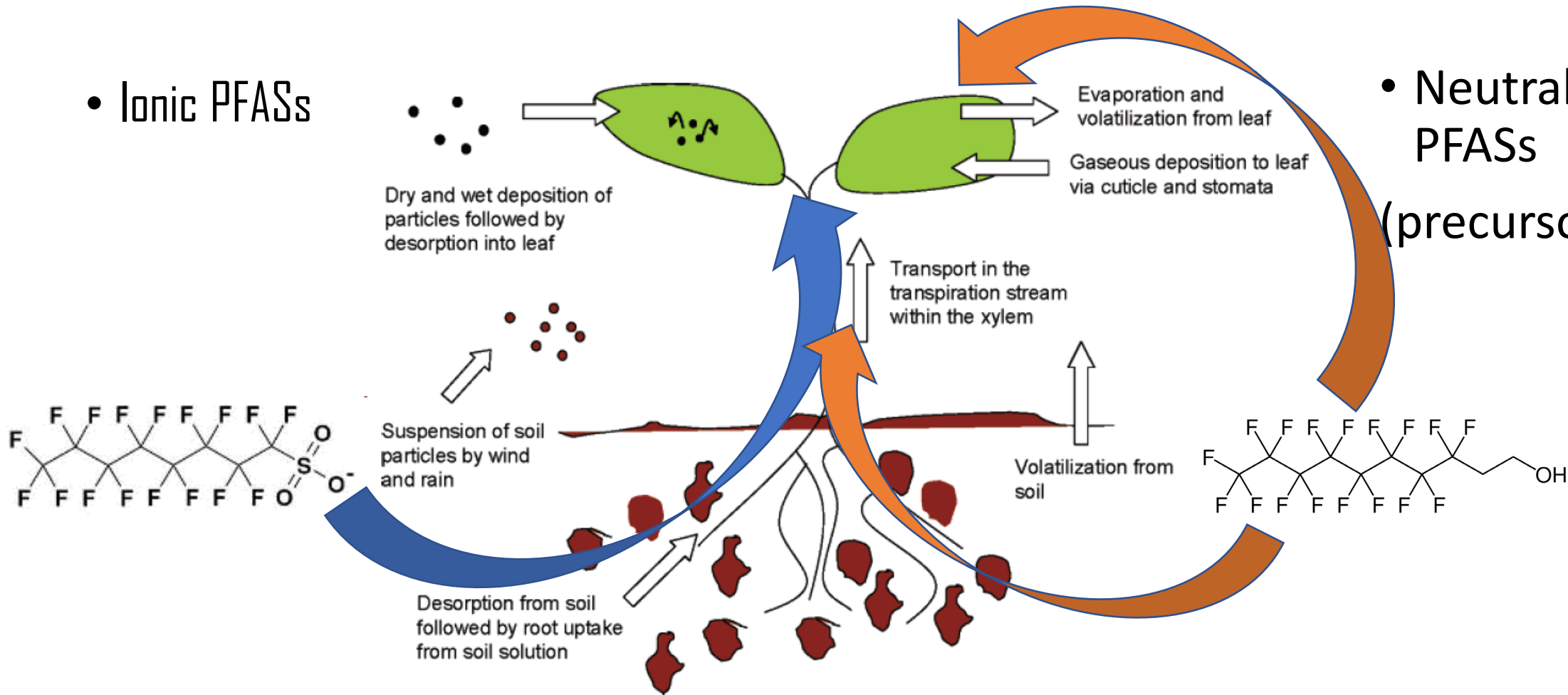
- Affects likely PFOA and PFOS, among others
- Evidence for importance of precursors in Arctic abiota
- And biota..



# PFAS in plants = PFIAntS?

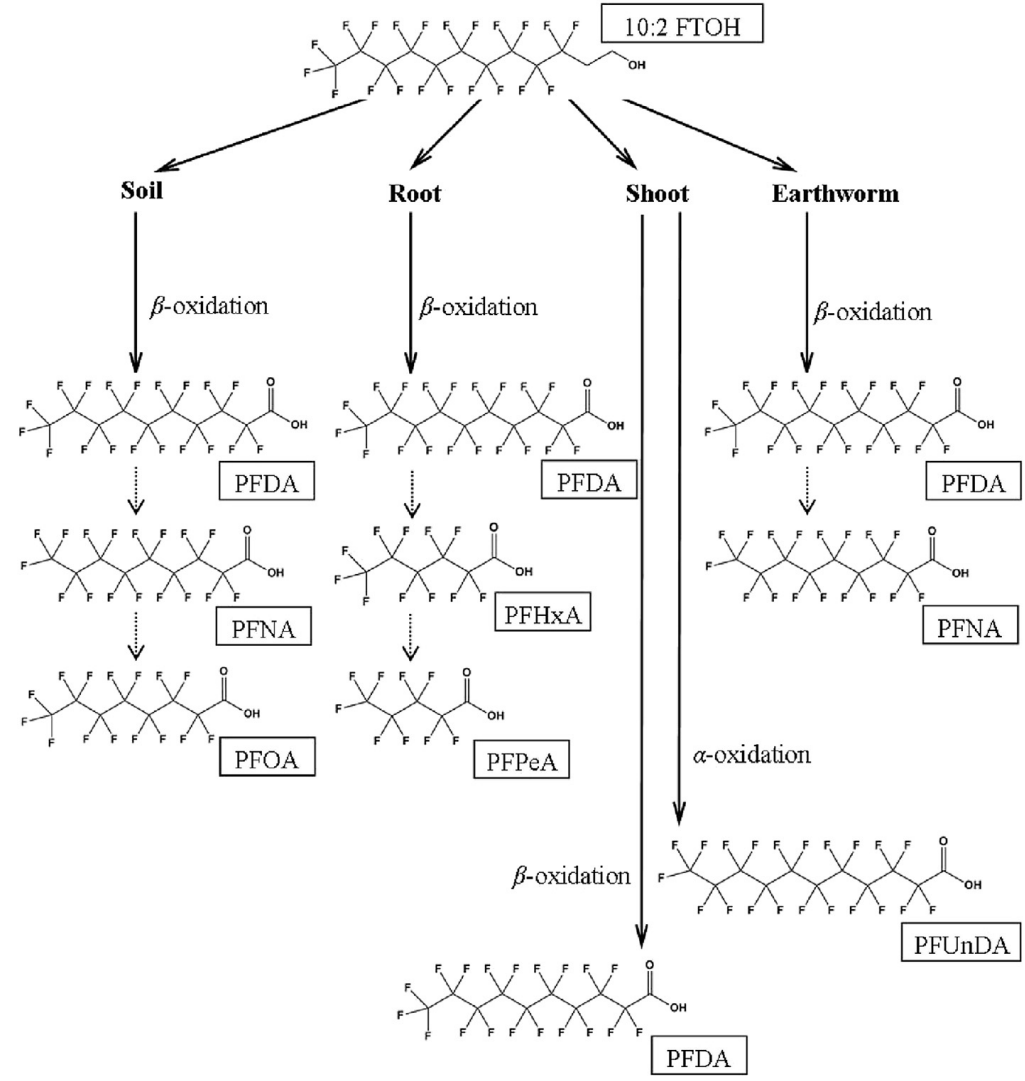
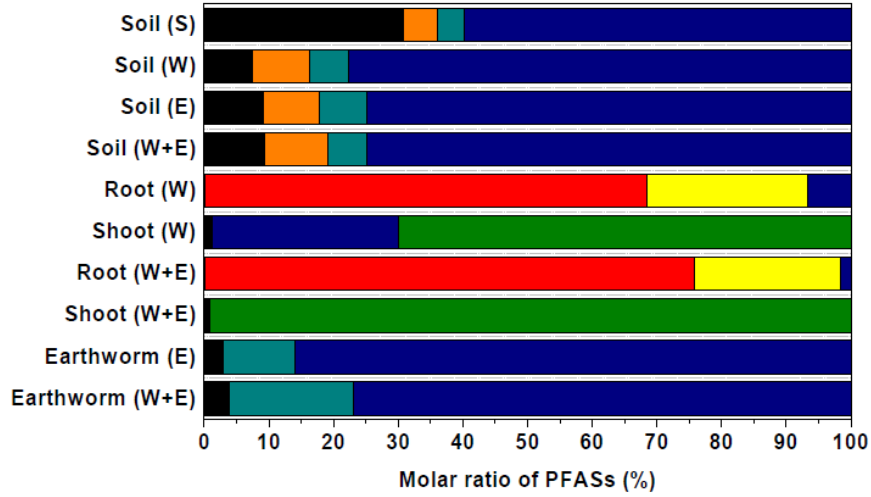
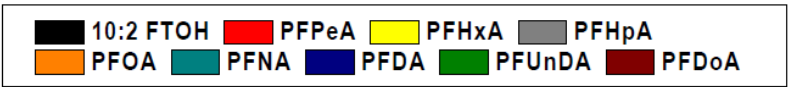
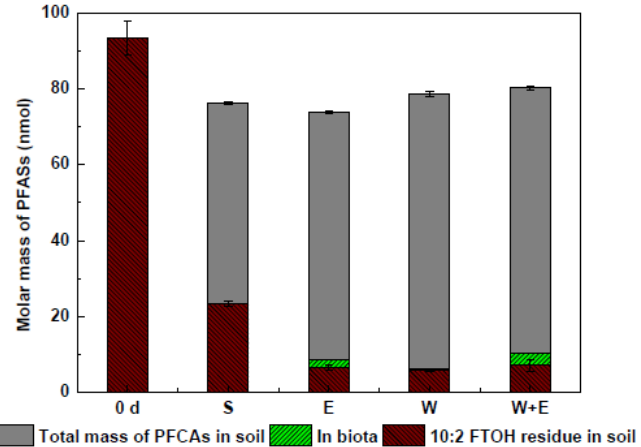
- Ionic PFASs

- Neutral PFASs (precursors)



(Collins et al., 2006)

# Chemists ♥ precursors – plants ⚡ them down



(Zhao and Zhu, 2017)



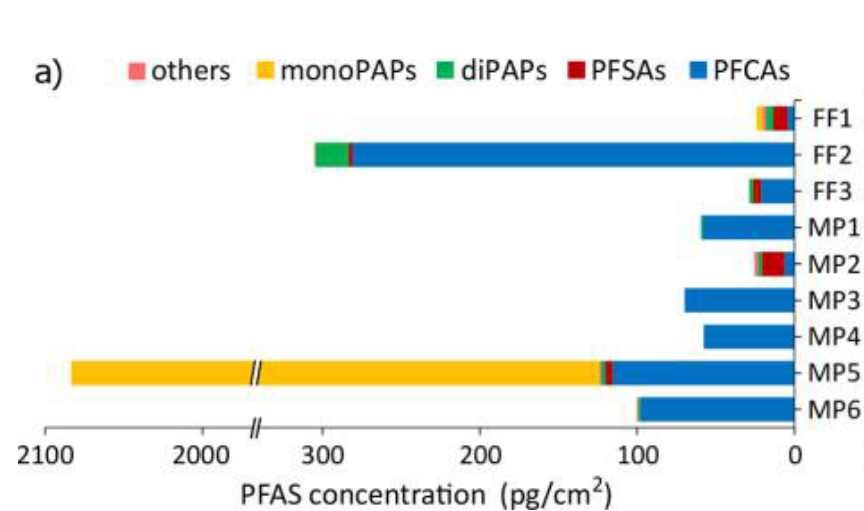
# Other breakdowns

	roots	leaves
8:2 FTOH	7:3 FTCA, 7:2 sFTOH, PFOA	PFOA
N-EtFOSAA	N-EtFOSAA, PFOS	PFOS, N-EtFOSAA
N-EtFOSA	PFOS, N-EtFOSA	PFOS, PFOSA
FOSA	PFOS, FOSA	FOSA, PFOS
6:2 FTSA	6:2 FTSA, PFHpA	6:2 FTSA, PFBA

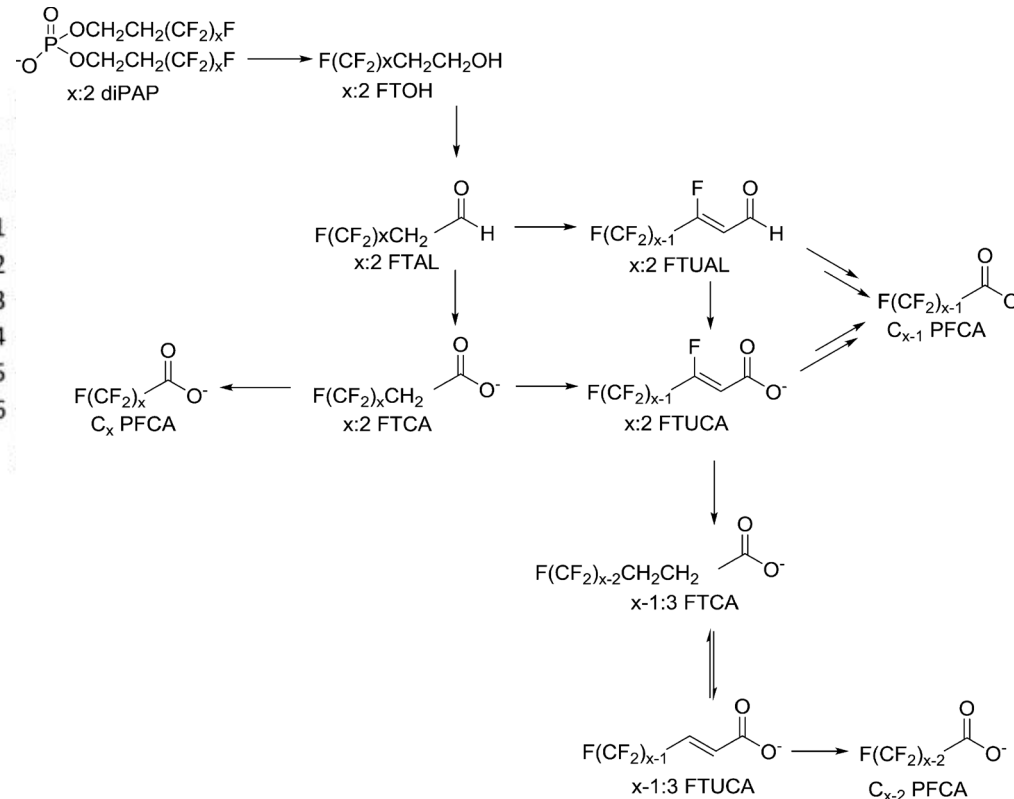
- Conversion efficiency depends on plant, precursor and time

(Wen et al., 2018; Zhang et al., 2016, Zhao, Zhou, et al., 2018, Zhao et al., 2019, Zhao, Liang, et al., 2018)

# Mono- & Di-polyfluoroalkylphosphates (food packaging)

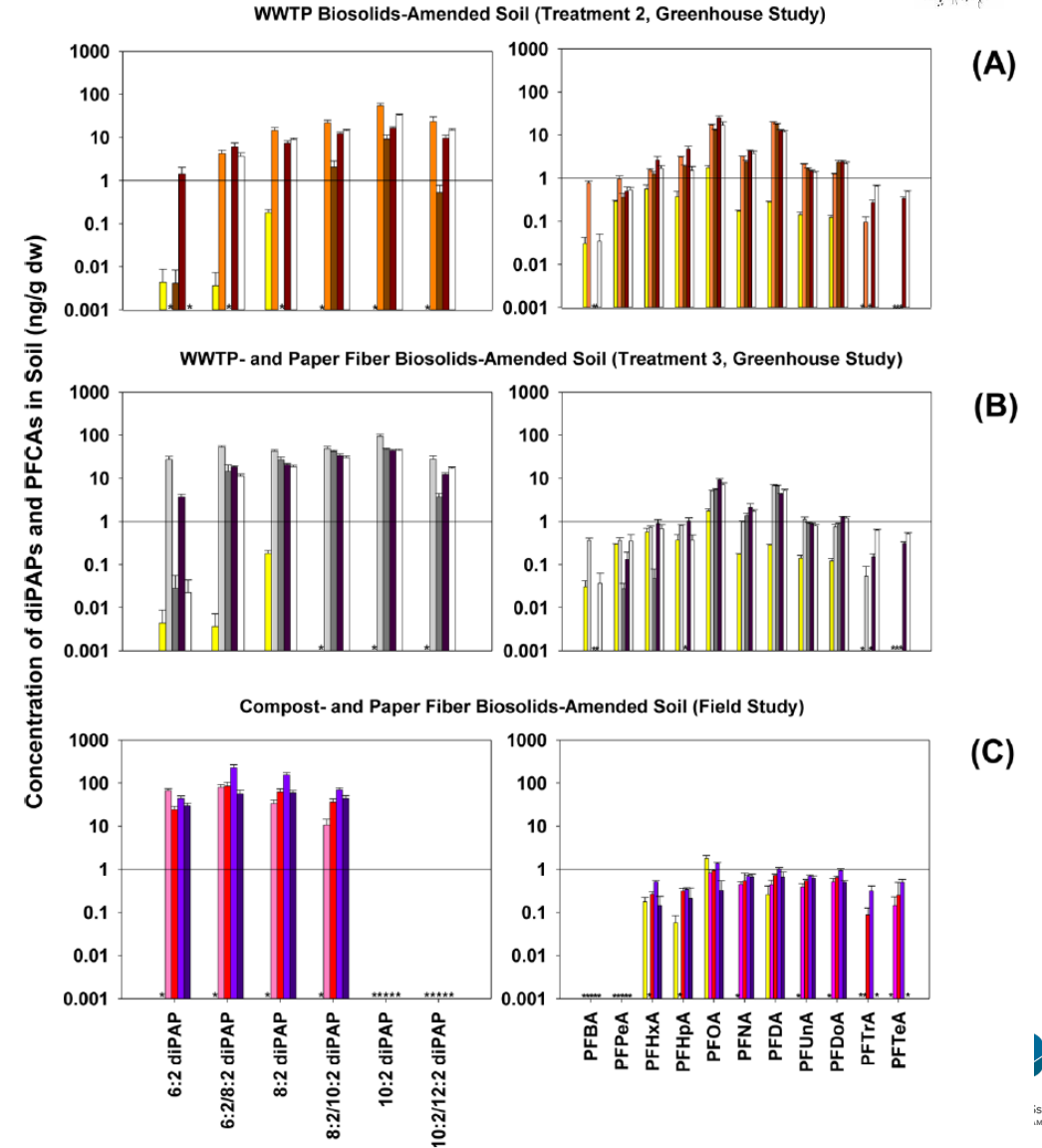
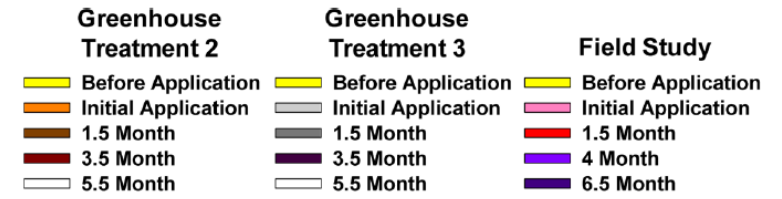
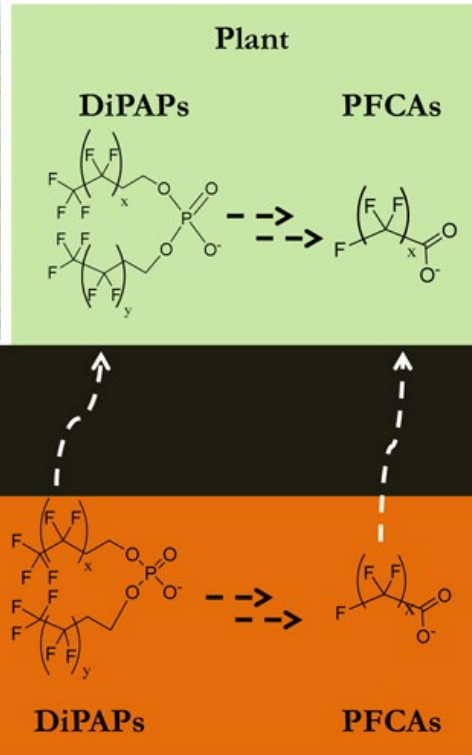


(Schultes et al, 2019)



(Rand and Mabury, 2014)

# Mono- and DiPAPs in plants



(Lee et al, 2014)

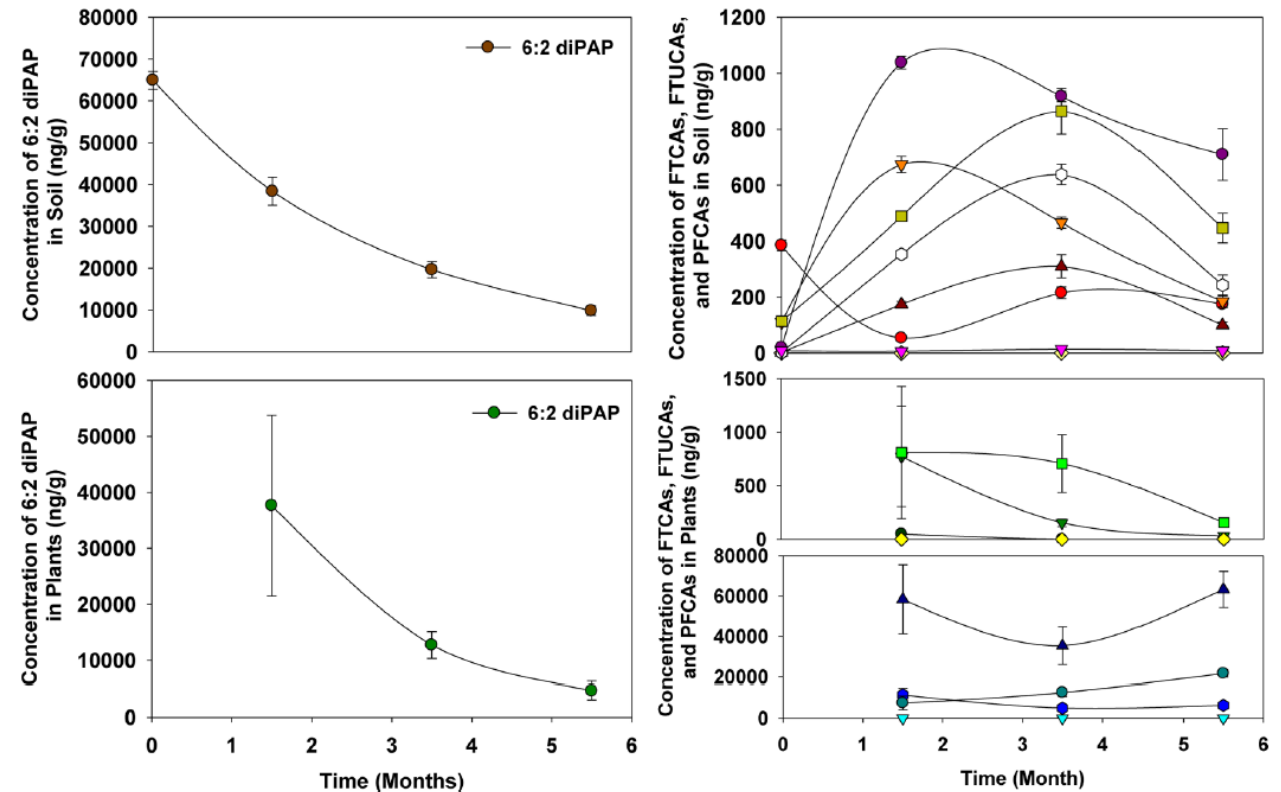
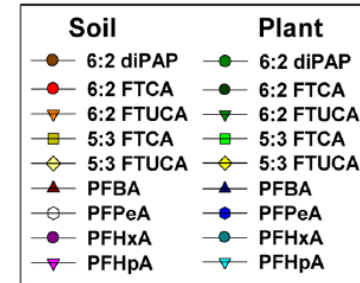


# Fate of spiked 6:2 DiPAP in soil/plant

$t_{1/2}$  of 2 months

Mostly in soil

1<sup>st</sup> hydrolysis to 6:2 FTOH  
 Then 6:2 FTCA, 6:2 FTUCA  
 -> PFHxA



(Lee et al, 2014)

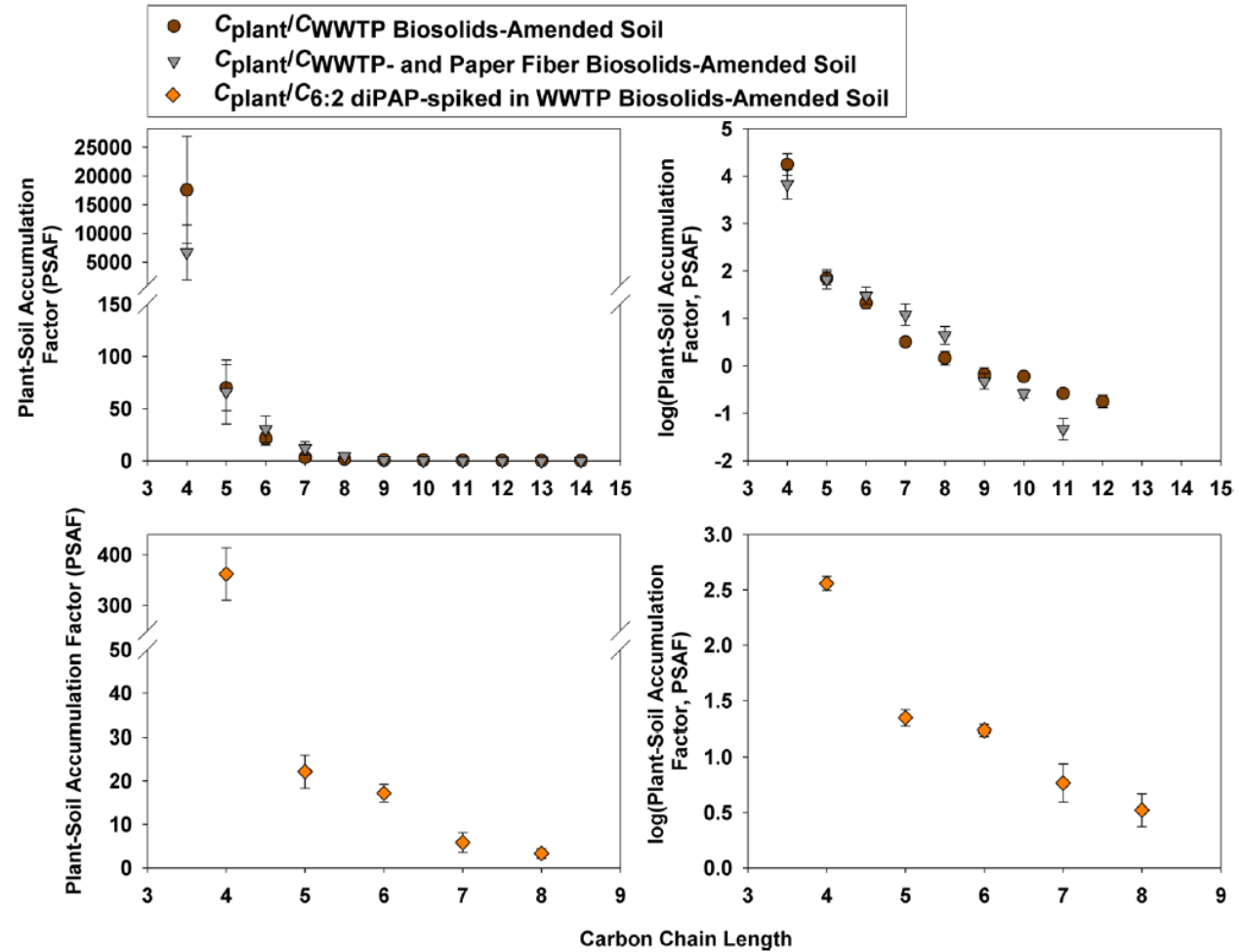
# Fate of spiked 6:2 DiPAP in soil/plant

$t_{1/2}$  of 2 months

Mostly in soil

1<sup>st</sup> hydrolysis to 6:2 FTOH

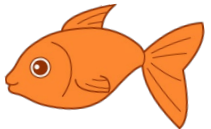
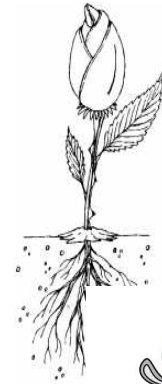
Then -> PFCAs



(Lee et al, 2014)

# Some conclusions and thoughts

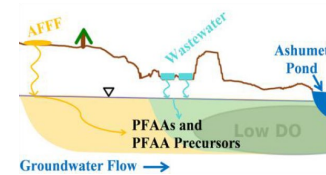
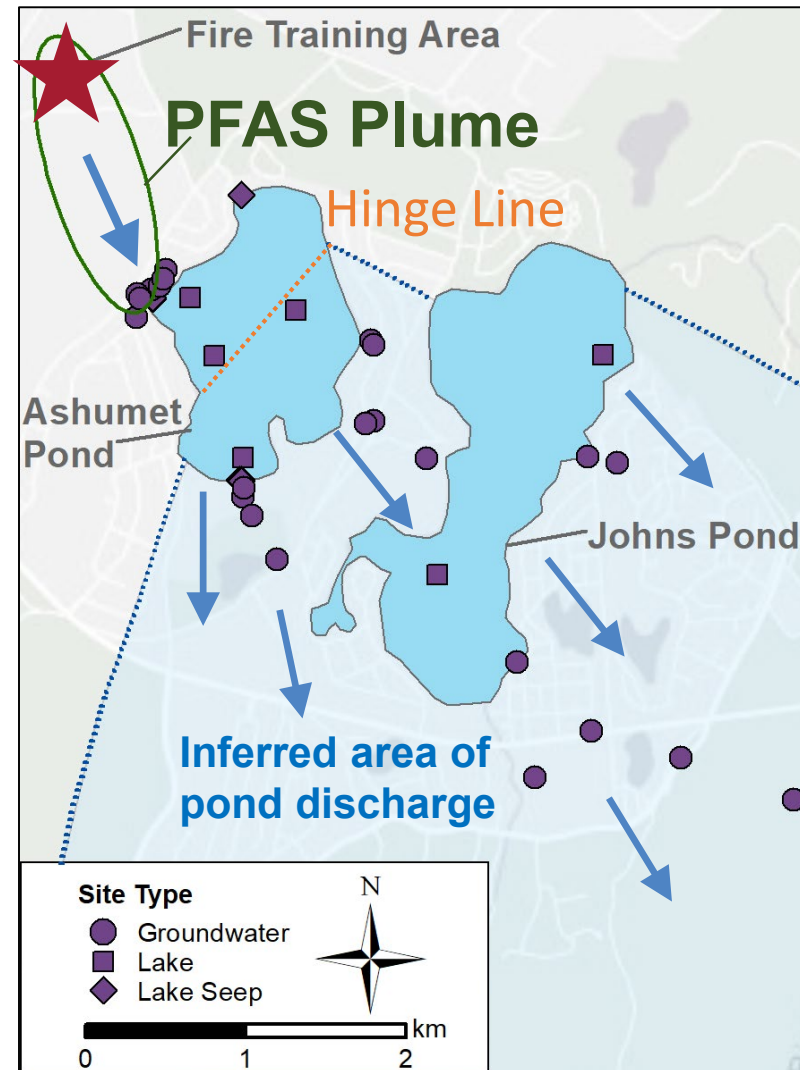
- Porewater concentrations?
- Plants return short-chain PFASs to foodwebs
- Potential opportunity for bio-removal of PFASs
- Environ. proteins and BC likely important for partitioning
- Knowledge gaps for terrestrial pathways
- Role of atmospheric precursors important (e.g., Arctic)
- Industrial vs marine signal for coastal biota
- The unknown bioaccumulating PFAS are of concern



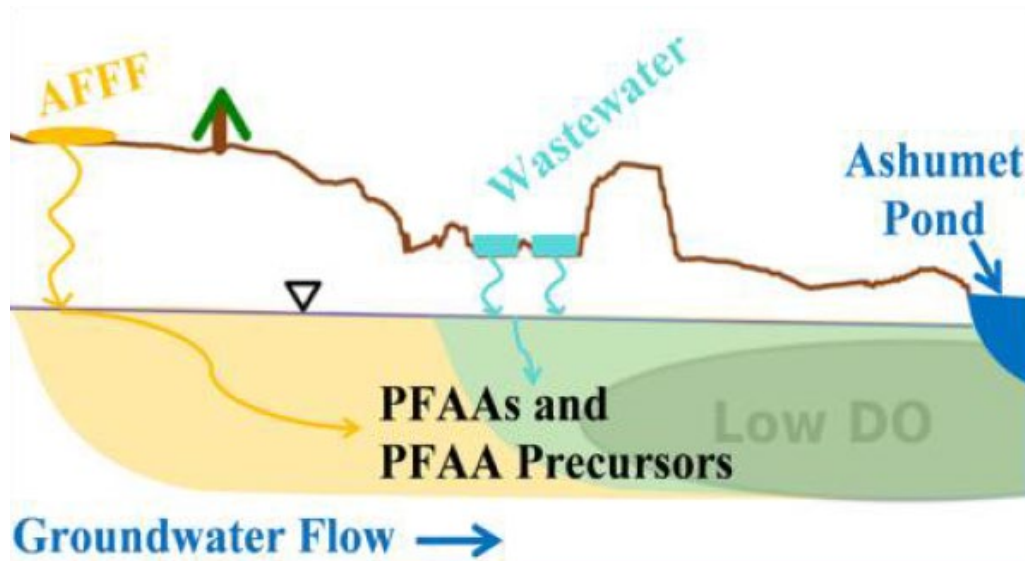
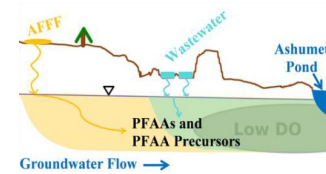
# AFFF precursors

# Investigating PFAS transport across groundwater/surface-water boundaries

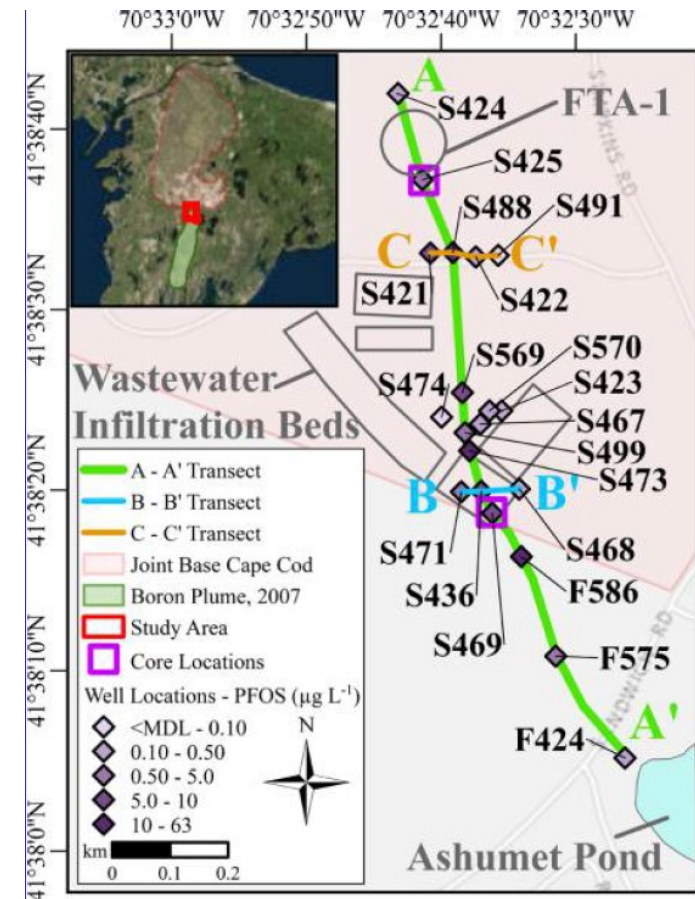
\*Interaction with a surface water body can cause significant dispersion of groundwater plumes



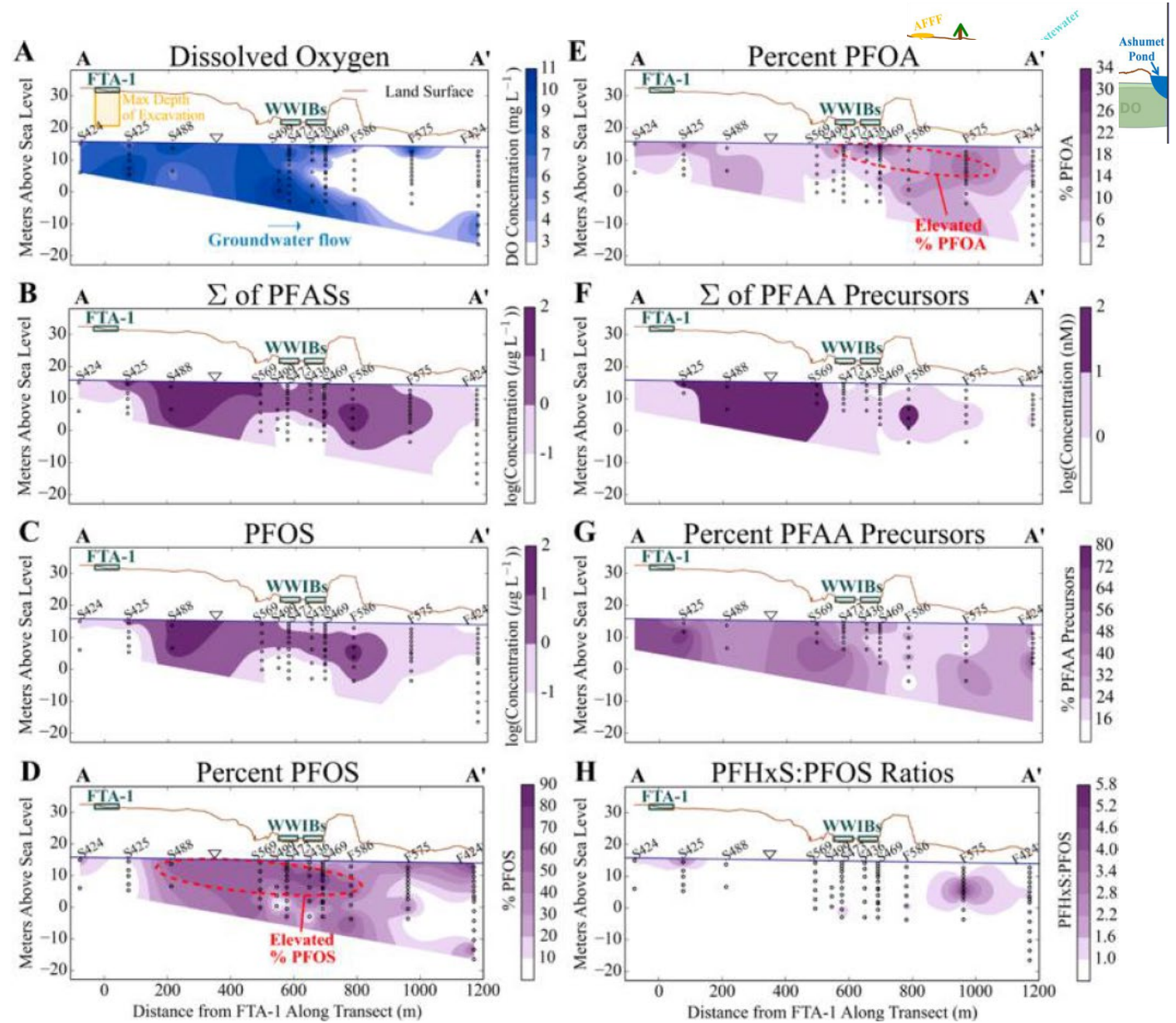
# Transport and fate of AFFF precursors down under



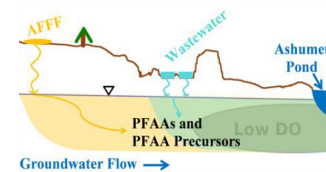
(Weber et al., 2017)



- AFFF plume from FTA-1
- 2<sup>nd</sup> source: WWTP effluent
- Precursors have not moved far
- Some evidence of transformation



(Weber et al., 2017)



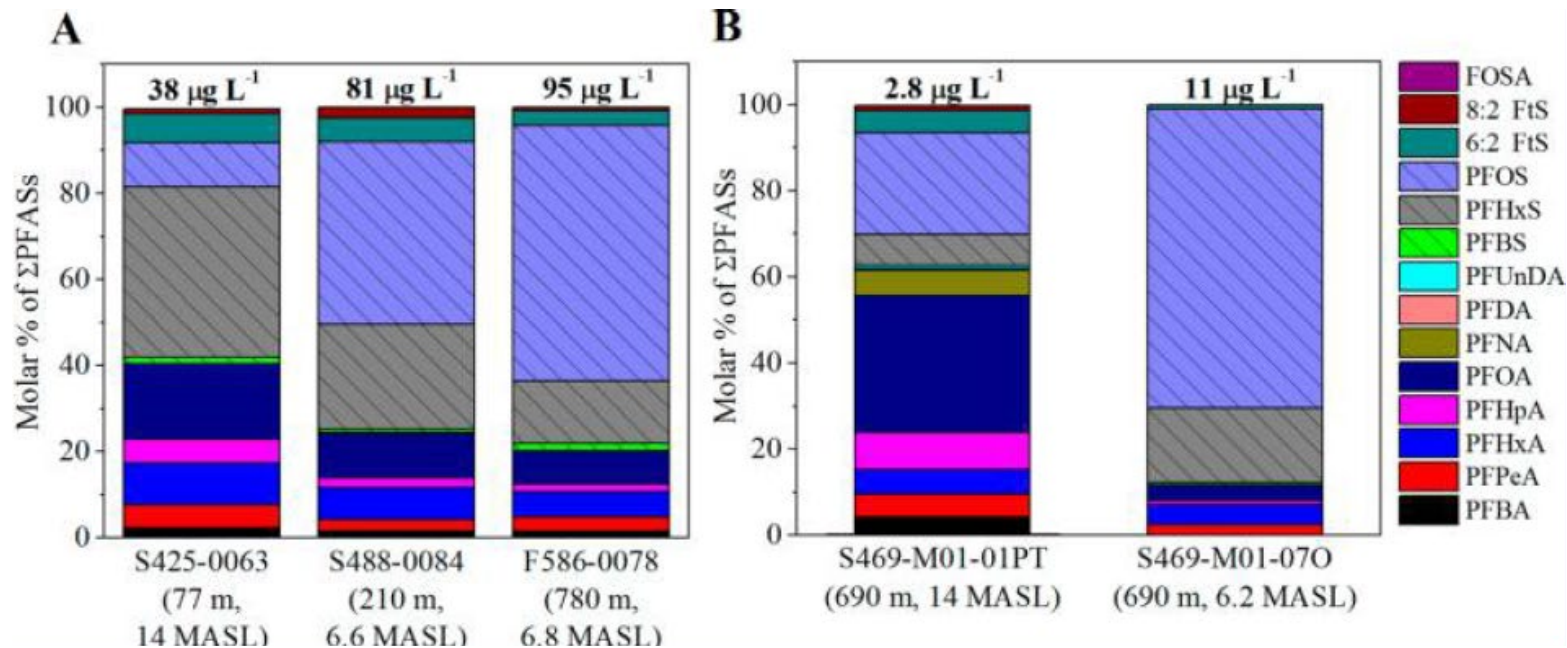
# The slow conversion of precursors

In AFFF plume

- Preponderance of PFHxS, PFHxA indicative of 6:2 FTS transformation

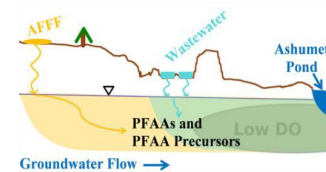
In WWTP recharge plume

- PFNA, PFOA, also from precursors



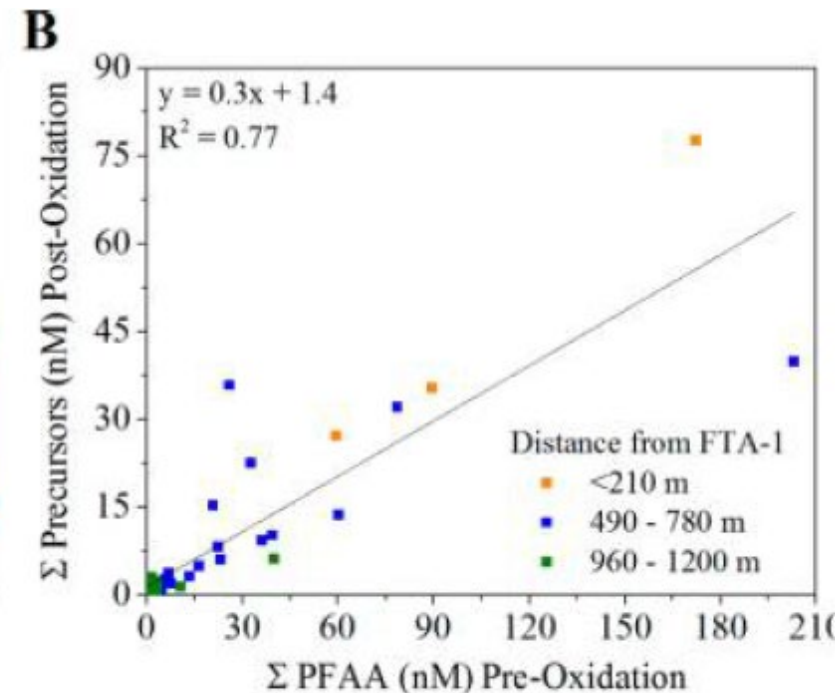
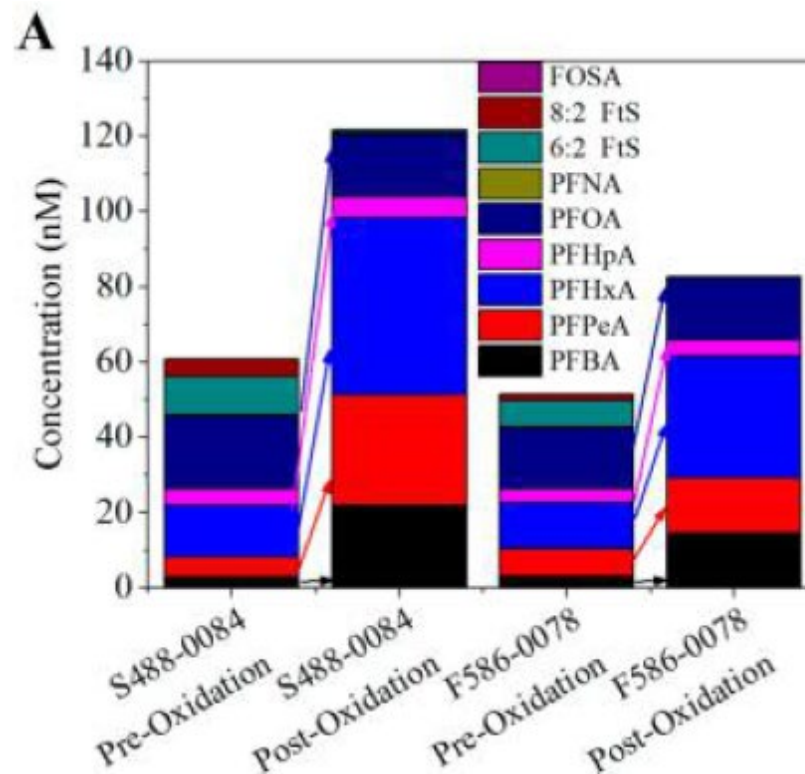
(Weber et al., 2017)





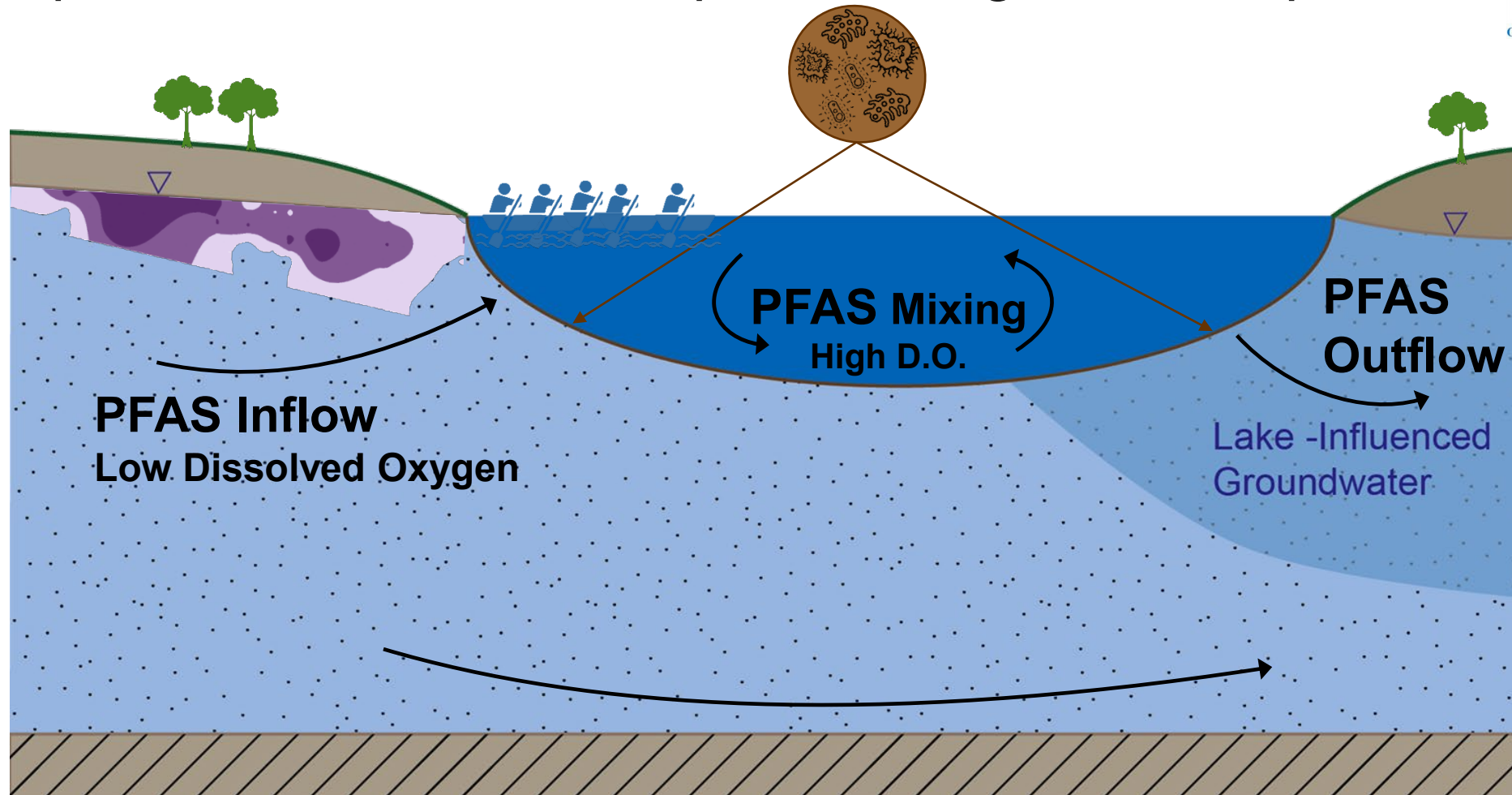
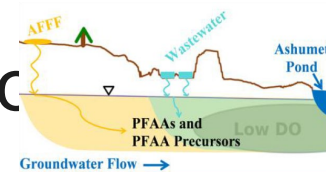
# Still more to be converted

- 6:2 FTS  
→ PFBA, PFPeA, and PFHxA
- 8:2 FTS  
→ PFPeA, PFHxA, PFHpA, and PFOA.



(Weber et al., 2017)

# Conceptual model – ideal setup to investigate PFAA precursors



Not to Scale



HARVARD  
John A. Paulson  
School of Engineering  
and Applied Sciences

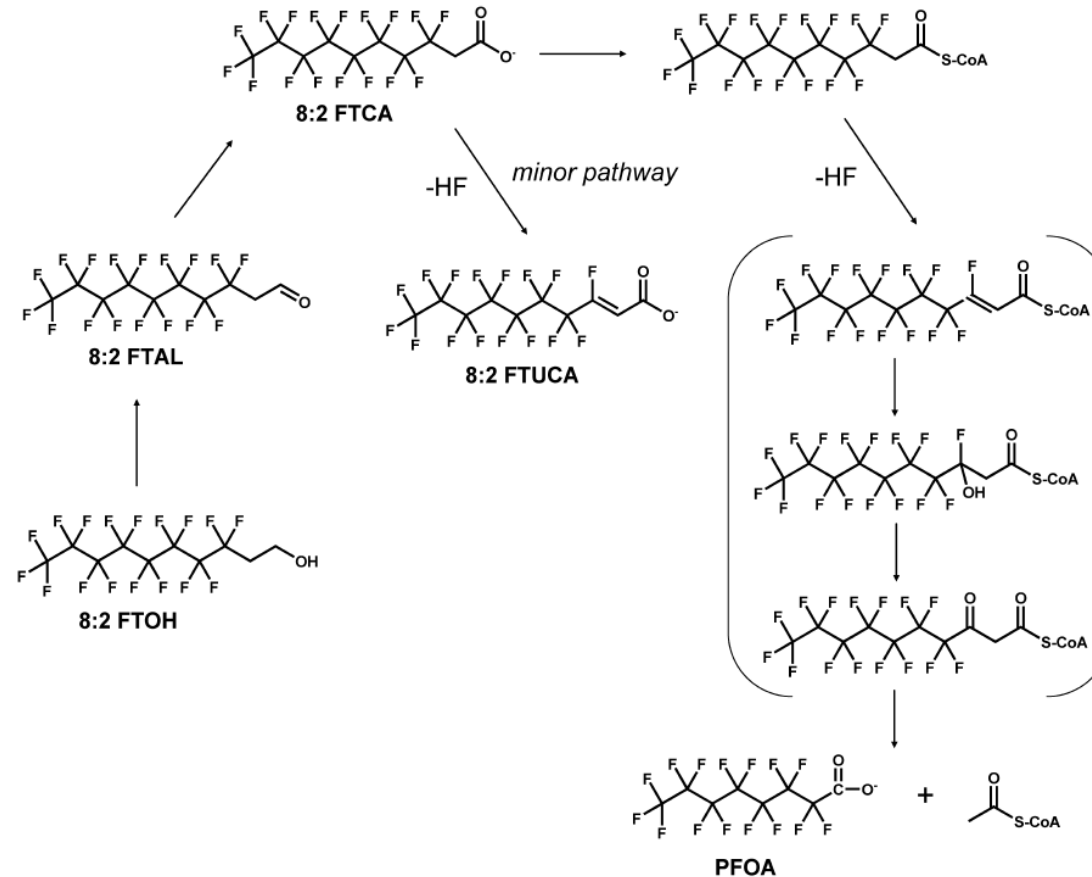
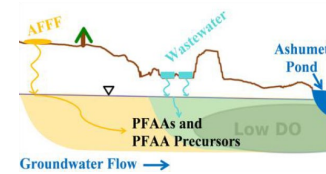


Preliminary Information – Subject to  
Revision. Not for Citation or Distribution.

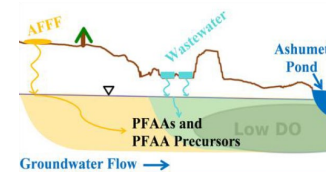


# Other microbial conversion

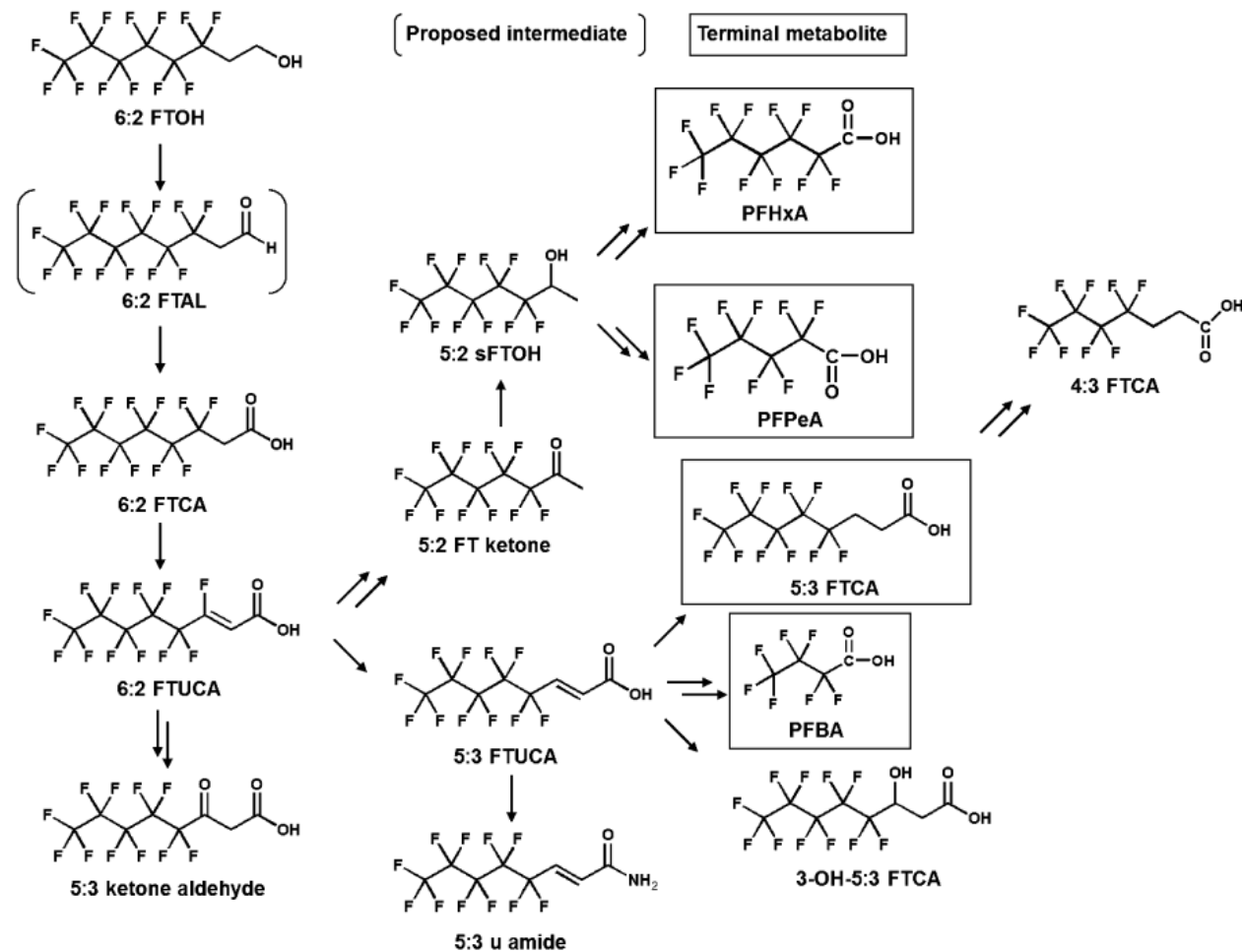
# Microbial breakdown of 8:2 FTOH



(Butt et al., 2014, after Dinglasan et al., )

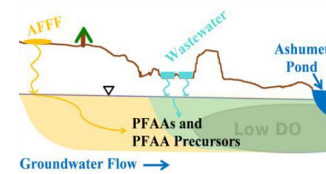


# Aerobic degradation of 6:2 FTOH

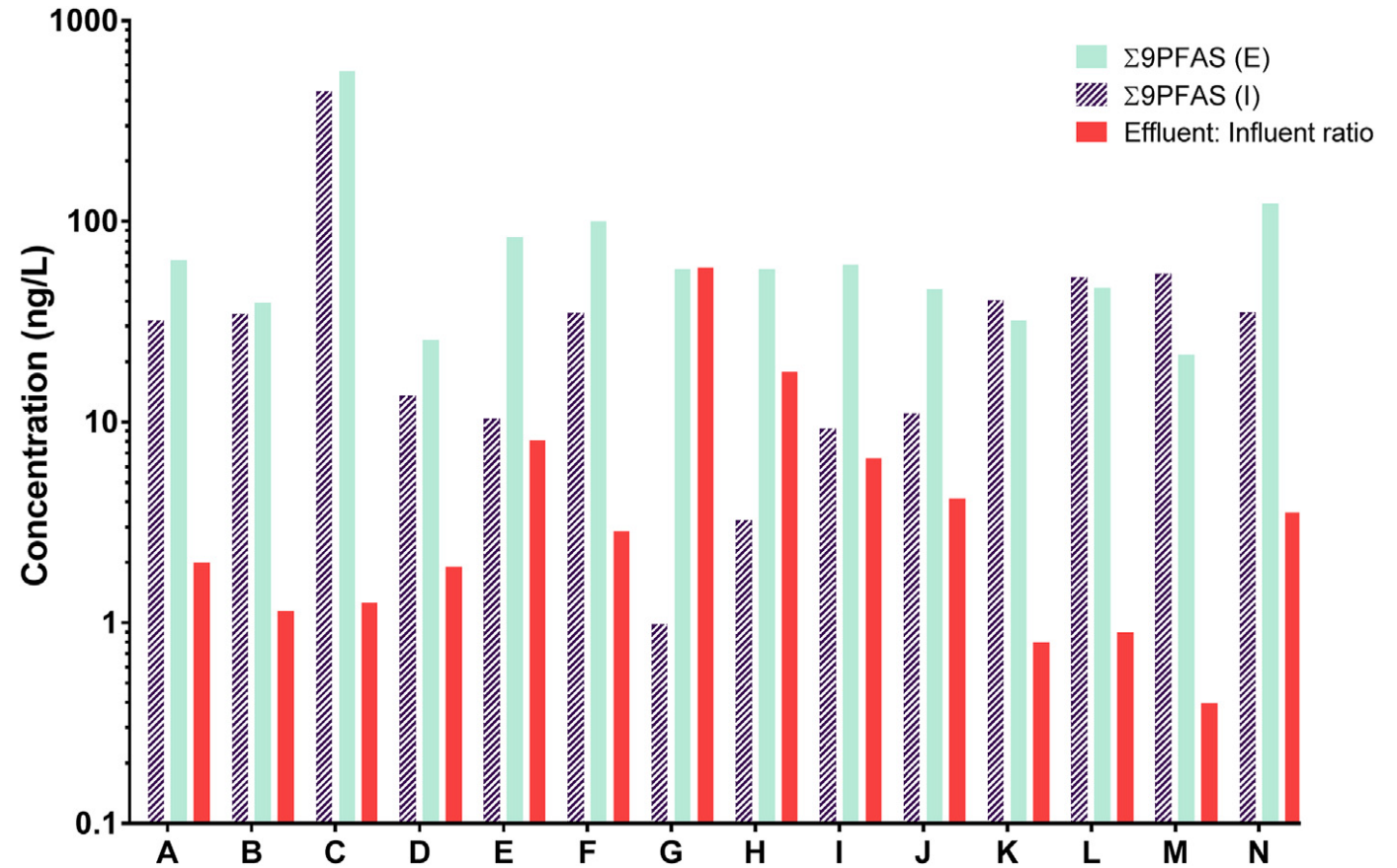
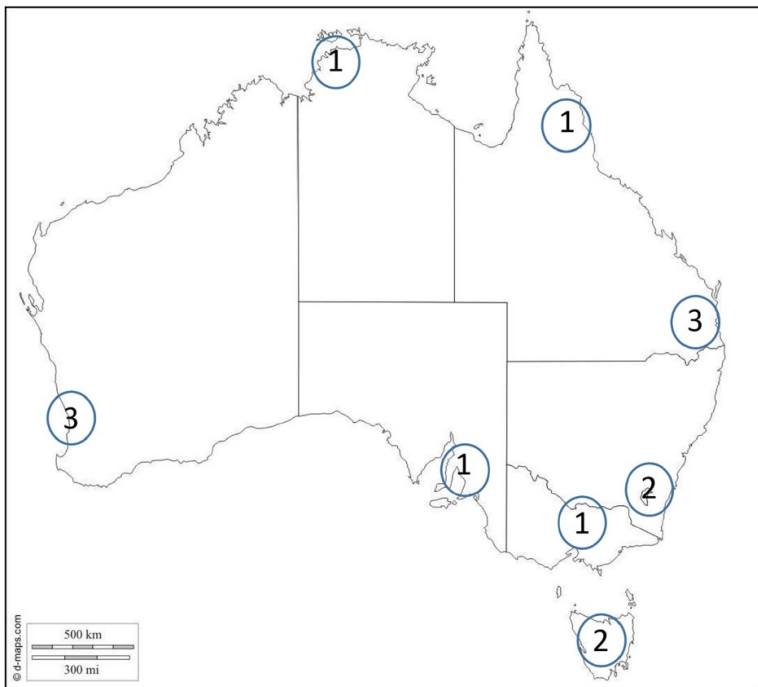


(Butt et al., 2014, after Liu et al., )

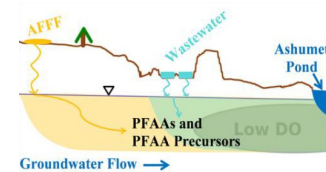
# 14 WWTPs sampled in 2014: Influent, effluent, biosolids



- Presence of PFASs down under  
– influent: effluent



(Gallen et al., 2018)



# WWTP down under

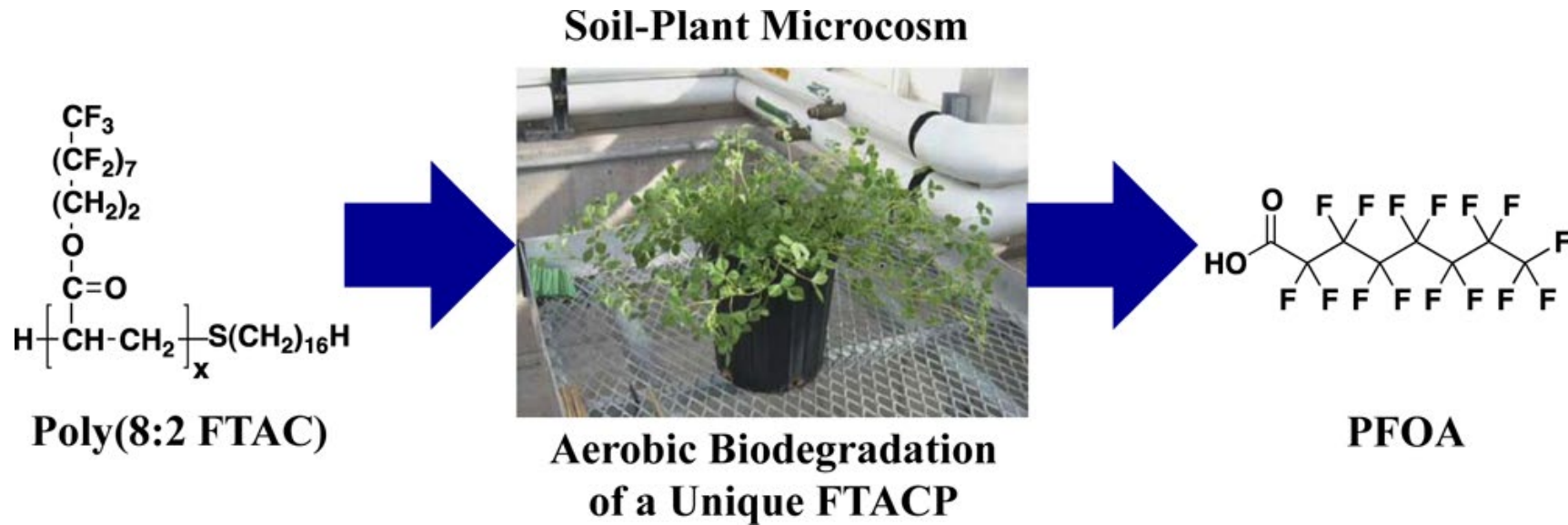
**Table 1**  
Summary results for influent, effluent and biosolids.

	PFHxA	PFHpA	PFNA	PFDA	PFUdA	PFDoDa	PFHxS	PFOA	PFOS	Σ9PFAS
<b>Influent</b>										
Mean	9.5	2.5	0.64	0.36	0.18	0.03	20	4.8	17	55
SD	15	3.3	2.0	0.68	0.31	0.10	54	6.9	35	117
Median	6.9	2.0	0.00	0.00	0.00	0.00	6.0	0.79	7.2	33
% detection	71	79	14	29	36	7.1	86	50	86	100
<b>Effluent</b>										
Mean	18	3.6	1.3	3.0	0.49	0.11	20	22	25	94
SD	9.2	1.2	0.73	2.3	0.19	0.21	3.5	15	7.3	29
Median	13	3.0	1.1	2.8	0.33	0.00	7.0	18	6.6	57
% detection	93	100	71	86	79	14	100	93	86	100
<b>Biosolids</b>										
Mean	0.50	0.26	1.1	17	1.2	5.7	0.11	6.5	25	45
SD	1.1	0.57	1.2	16	1.0	4.6	0.26	8.1	31	50
Median	0.00	0.00	1.0	11	1.4	5.8	0.00	0.0	11	40
% detection	27	18	64	100	73	82	18	45	91	100

- Strongest increases for PFCAs, much less so for PFSAs
- PFDA (9x), PFOA (4x), PFUdA, PFDoDa (3x), PFNA, PFHxA (2x)

(Gallen et al., 2018)

# Potential for side-chained fluoropolymer to liberate precursors



(Rankin et al., 2014)



# Some summarizing thoughts

- Precursors are everywhere.
- 8:2 FTOH and 10:2 FTOH with global presence, and caused baseline contamination of Arctic (and global?) air, snow, water.
- FOSA likely implied in contamination and sharp decrease of PFOS in Arctic region, including trends in biota
- Atmosphere, plants, microbes breakdown precursors, with different endproducts
- WWTP efficient transformers of precursors; less obvious for groundwater
- Much greater indoor concentrations, and associated human exposure

Please Hold the Date:

**FLUOROS 2020**

Oct 13-16, Providence (RI)

**Questions?**

THE  
UNIVERSITY  
OF RHODE ISLAND



SCHOOL OF PUBLIC HEALTH  
Department of Environmental Health



STEEP is funded under award number P42ES027706.  
More information about STEEP is available at: [www.uri.edu/steep/](http://www.uri.edu/steep/)

